A SENSITIVE METHOD FOR QUANTITATION OF HYDRAZINE IN MAINSTREAM TOBACCO SMOKE

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Objective

To develop a sensitive method for quantitative analysis of trace levels of hydrazine in mainstream tobacco smoke.

Challenges:

- sensitivity
- selectivity
- robustness

Analytical Background

Hydrazine is a small, polar, basic and strongly reducing agent.

Urine Plasma Serum





Conventional methodology for analysis of hydrazine: "Schiff base" formation

Analytical Challenges

Previous approaches for analysis of hydrazine in mainstream tobacco smoke:

Ref.	Sample Generation/Extraction	Sample Preparation	Analysis	Detection Limit	[Hydrazine] (ng/cig.)
Hoffmann (1974)	. 20 <u>non filter</u> cig. / "ISO" smoking . trapping solution (aqu. buffer/PFBA) . no filter pad	. Silica gel plate (2X) . TLC (Al ₂ O ₃ plate) . Extract with ether	GC/ECD	0.1 (ng/ cig.)	31
Plunkett (2002)	. puff-by-puff measuring of a flowing stream of smoke	none	IR-TDL	0.4 (ppmv)	ND

Hoffmann et al., *Anal. Chem.*, 46(7), 885-889, 1974 Plunkett et al. *Spectrochimica Acta Part* A, 58, 2505-2517, 2002

Analysis of hydrazine in tobacco smoke requires a quantitative isolation procedure to prevent its loss due to:

- its oxidation with other reactive smoke constituents
- poor stability of hydrazine derivatized products

Analytical Approach

Derivatization Reagent

Compare derivatization efficiencies of benzaldehyde derivatives

Internal Standard

Investigate and Select the Internal Standard

Analysis

Separation and Detection Methods

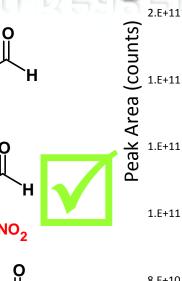
Trapping Mechanism

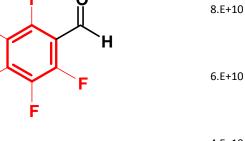
• Establish a quantitative trapping /extraction procedure from smoke (i.e. effect of glass fibre filter pad, Impinging volume etc...)

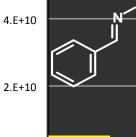
Derivatization Reagent

Benzaldehyde

Pentafluorobenzaldehyde

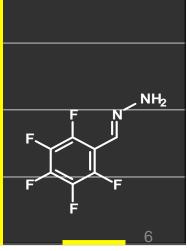




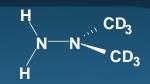


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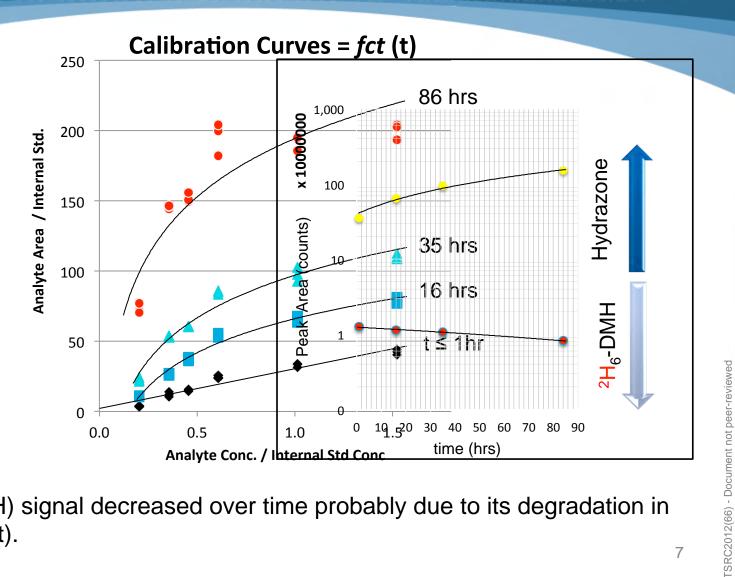
NH₂



NO₂



- Selection of Internal Standard: ²H₆-Dimethylhydrazine

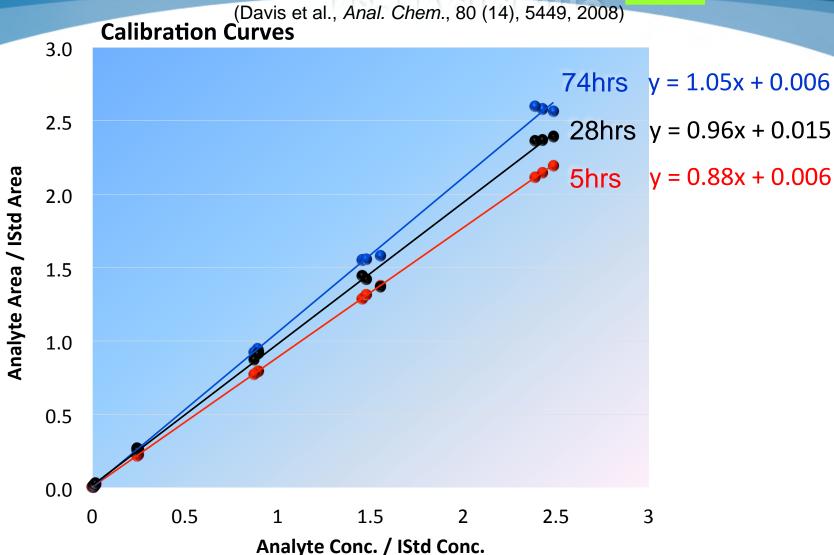


Internal Std (DMH) signal decreased over time probably due to its degradation in methanol (solvent).

Method Development - Internal Standard:

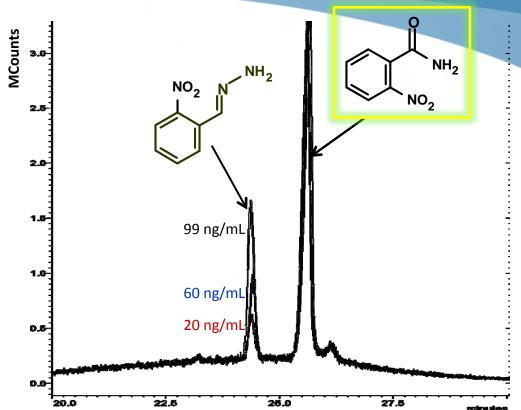
¹⁵N₂-Hydrazine





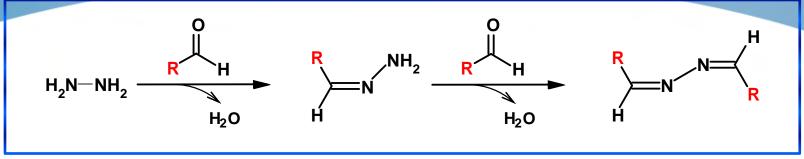
- Analysis Method: GC-MS

GC column	Wax	
	30 m x 0.25 mm x 0.25 μm	
Injector Temp.	260°C	
Column Temp.	55°C, hold for 2 minutes	
	15°C / minute to 150°C,	
	hold for 25 minutes	
	10°C / minute to 240°C,	
	hold for 2 minutes	
Constant Flow	1.4 mL/min	



2-NBA undergoes an adverse reaction to form 2-nitrobenzylamide which generates an ion at m/z 166 (isobaric product ions) in the mass spec. source.

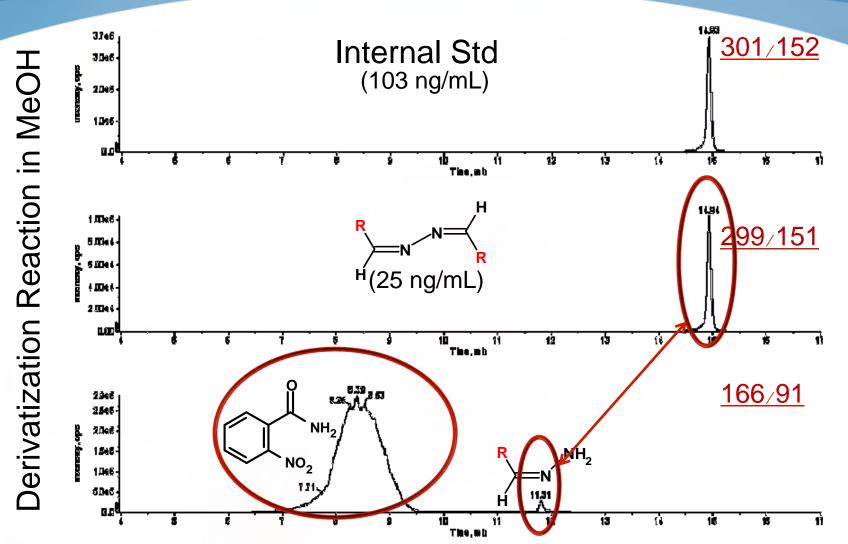
- Analysis Method: LC-MS/MS



MRM Transition (<i>m/z</i>):	166 / 91	299/151
, ,		299 / 78
		301 / 152 (ISTD)
		301 / 105 (ISTD)

Ionization / Mode:	ESI/MRM
Polarity:	Positive
IonSpray Voltage:	5500 V
Source Temperature:	400°C
Curtain Gas Flow:	12 mL/minute
CID Gas Flow:	12 mL/minute
Nebulizing Gas:	9 mL/minute

- Analysis Method: LC-MS (Chromatogram)



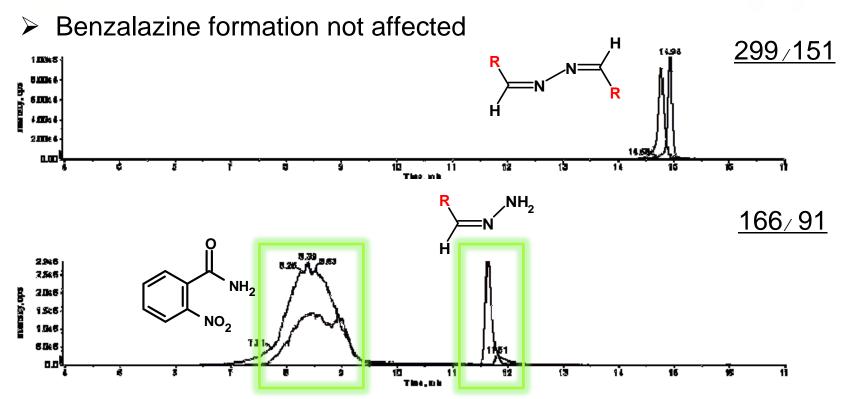
- Analysis Method: LC-MS

MeOH

VS.

Na₂HPO₄:MeOH (55:45, v/v)





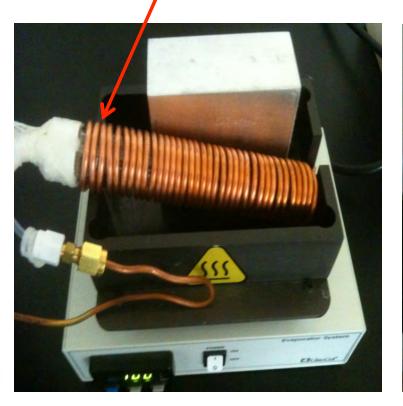
Hydrazone formation significantly favoured in acidified solvent

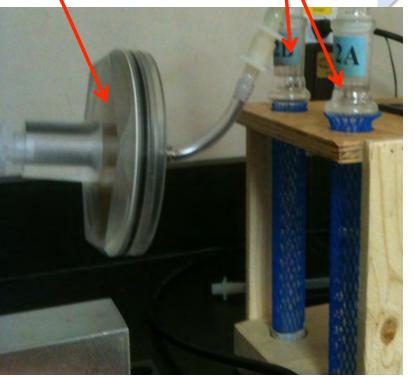
- Evaluating Trapping Efficiency (Apparatus)

Permeation Tube

Pad Holder

Trapping Solution

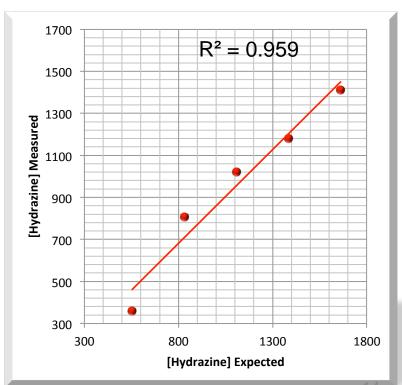




- Evaluating Trapping Efficiency (Permeation)

[Hydrazine] =
$$\frac{P_{\times} K_0}{F_{\times} V} \times t$$

- P permeation rate (Temp. dependant)
- K₀ molar constant (0.699)
- F Carrier gas (N₂) flow rate (mL/minute)
- V trapping volume (mL)
- t sampling time (min)



- Trapping Efficiency of Hydrazine Vapour

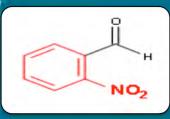
Parameters	Values	Comment
Temperature (Permeation Tube)	~ 70°C	Measurement after equilibration period (4 hrs)
Trap solution Volume (V)	2 X 40 mL	Wash bottles connected in series
Sampling Period (t)	15 min.	Hydrazine [lawing]drawn through the system
Recovery % Calculation		$\frac{measurea}{[Azine]} \times 100$ $ref.(trap)$

Trials Description			
Reference (<u>no Pad</u>)			
Blank Pad			
Smoked Pad (TPM)			
DNPH-treated Pad			

[Azine] Measured				
Filter Pad	Filter Pad Trap A (ng/			
(ng/mL)	mL)	mL)		
-	637 (ref.)	nd		
546	96	nd		
621	14	nd		
766	6	nd		

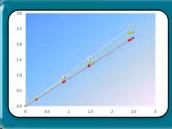
Recovery			
Filter Pad	Trap A		
-	100 %		
86%	15%		
98%	2%		
120%	1%		

Summary of Findings



Reagent / Trapping Solvent

- 2-Nitrobenzaldehyde (10g/L)
- Aqueous Phosphate Buffer Solution (PBS:MeOH, 55/45)



Internal Std / Solvent

- Isotope dilution Calibration using ¹⁵N₂-Hydrazine
- Calibration Stds prepared in 0.1 % CH₃COOH solution



Detection

- LC : Core-Shell PFP phase; acidified buffer 0.1% CH₃COOH
- ESI⁺-MS/MS (MRM)



Trapping Mechanism (Hydrazine vapour)

- Volume of Trapping solution: 40mL
- Hydrazine mostly present in PP

Methodology

Sample Generation

- 10 cigarettes smoked
- 92mm glass fiber filter disc
- 40mL of impinging solution (PO⁴⁻: MeOH; 2-NB; ISTD)

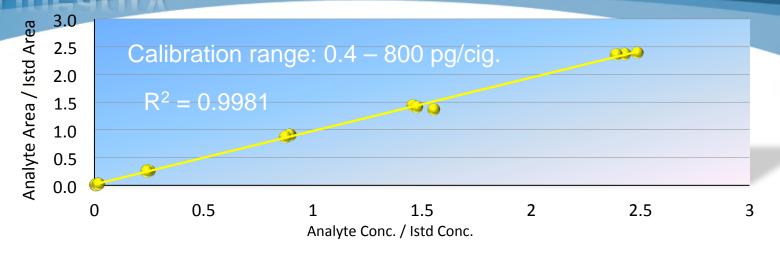
Sample Extraction

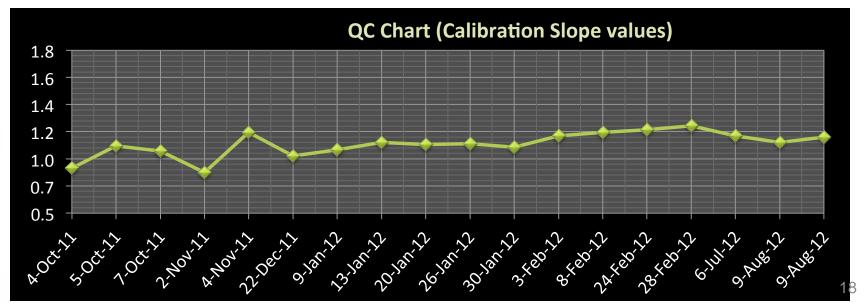
- Extract filter pad with impinger solution (40 mL)
- Dilute with fresh derivatization solution (40 mL)
- Incubate extract for 30 minutes at 35°C
- Shake extract for 30 minutes
- Centrifuge at 10,000g for 5 minutes

Sample Analysis LC-ESI-MS/MS operating under MRM mode

Analytical Performances

- Linearity





Analytical Performances- Sensitivity (Instrument)

Injection Date/Time	[Hydrazine] (pg/cig.)	Recovered as % of Expected
Day 1	603	96%
Day 1	578	92%
Day 1	594	94%
Day 1	628	99%
Day 2	647	102%
Day 2	628	99%
Day 2	603	95%
Day 2	686	109%
Day 2	651	103%
Day 3	681	108%
Day 3	694	110%
Day 3	669	106%
Average	639	101
Std Dev	39	6
RSD (%)	6.1	6.2

Limit of Detection Limit of Quantitation

0.116 ng/cig.

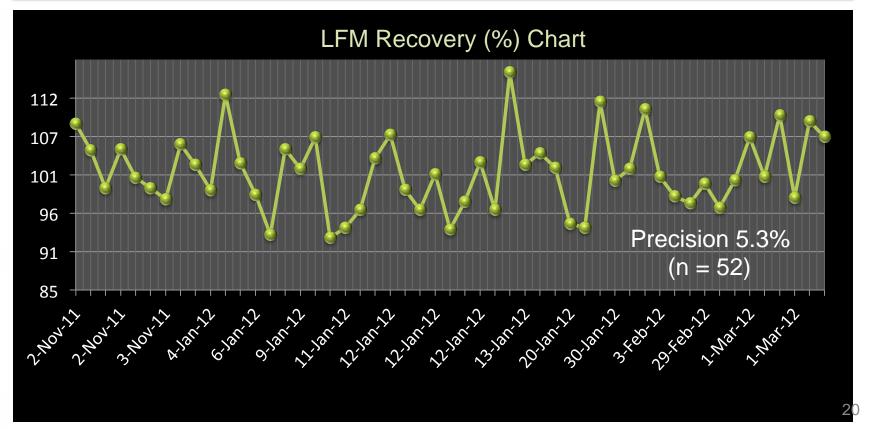
0.388 ng/cig.

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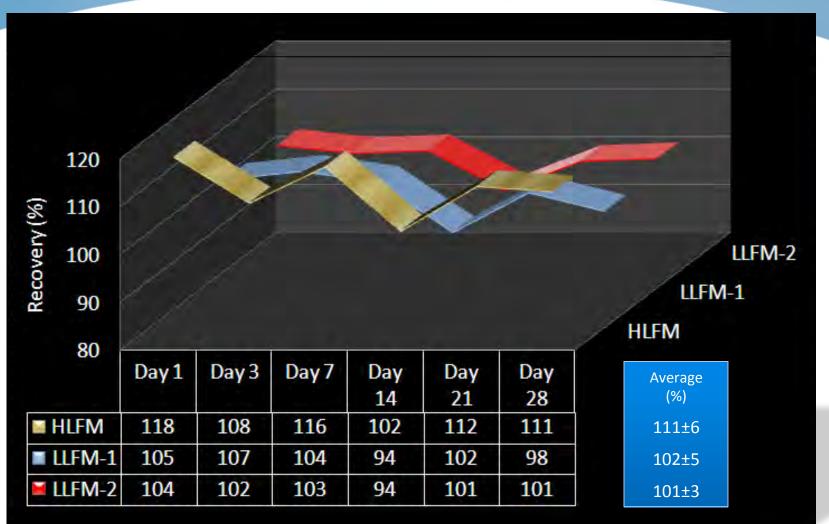
Analytical Performances

- Method Recovery

Fortification Range (ng/mL)	Recovery (%)	Precision (n = 16)
60 - 120	85 - 110	6.2 %



Analytical Performances- Stability of Extracted Samples



Conclusion

- A quantitative, sensitive and reliable method for analysis of hydrazine in tobacco smoke was developed.
- Scope of application: Quantitation of hydrazine in tobacco smoke (i.e. MS/SS) generated under both "ISO" and "intense" smoking regimens.

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