

# Pesticide Fate in Soil and Plants

*Travis W. Gannon, Ph.D*



# What We'll Talk About

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- Why is pesticide fate important?
  - Pesticide behavior in soil
- Systemic nature of pesticides

# Pesticide Fate

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## Why is it important???

Preserve tools (pesticides) today,  
while providing information required  
for the implementation of best  
application practices in the future

# Pesticide Fate

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## Why is it important???

- Keep the right pesticides on the market
- Residue management
- Improve comprehensive IPM plans
- Reduce potential for adverse effects on off-target areas or species
  - off-target movement = angry neighbors or health issues*





# Pesticide Environmental Fate

Photodecomposition

Drift

Runoff

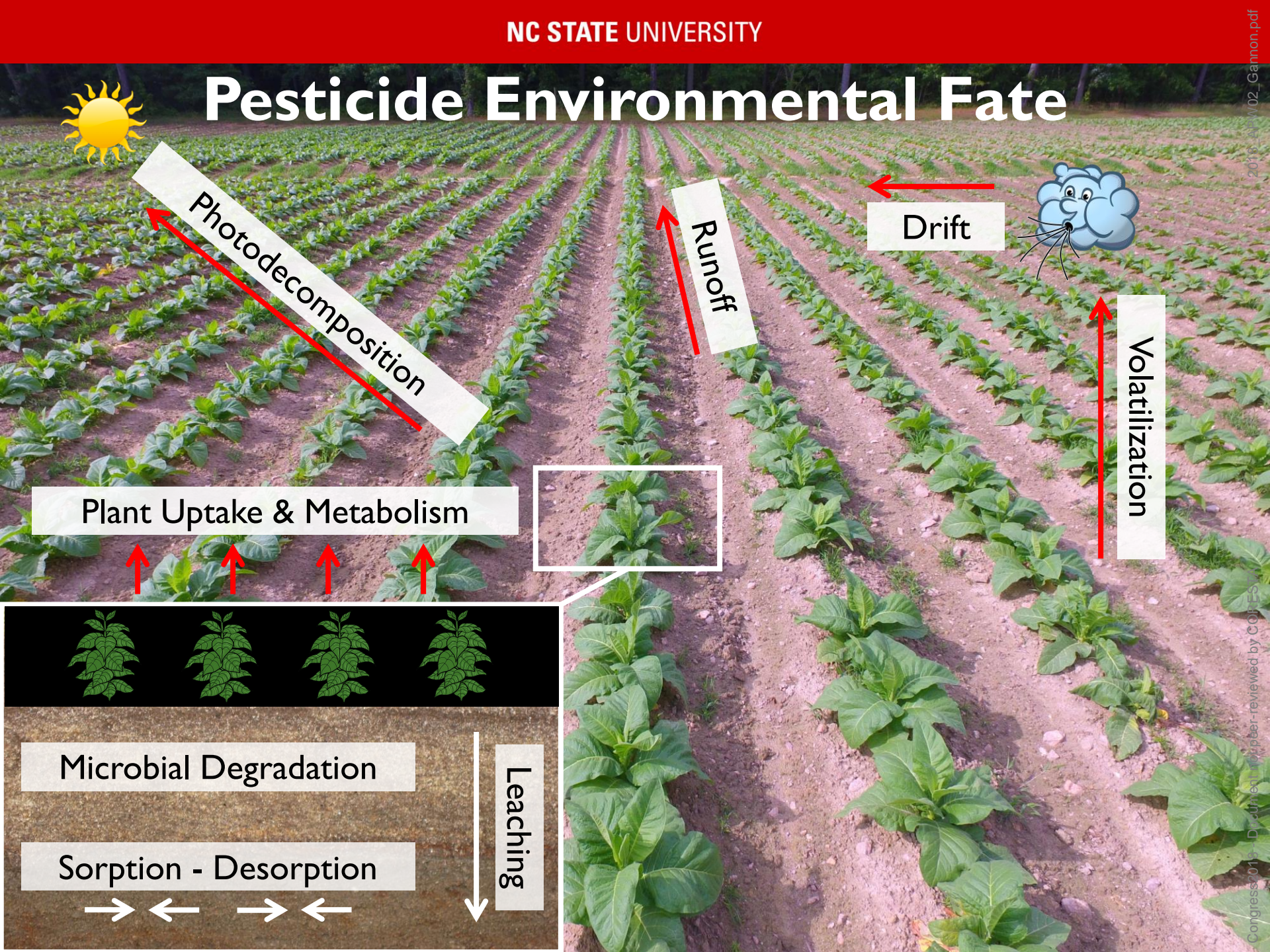
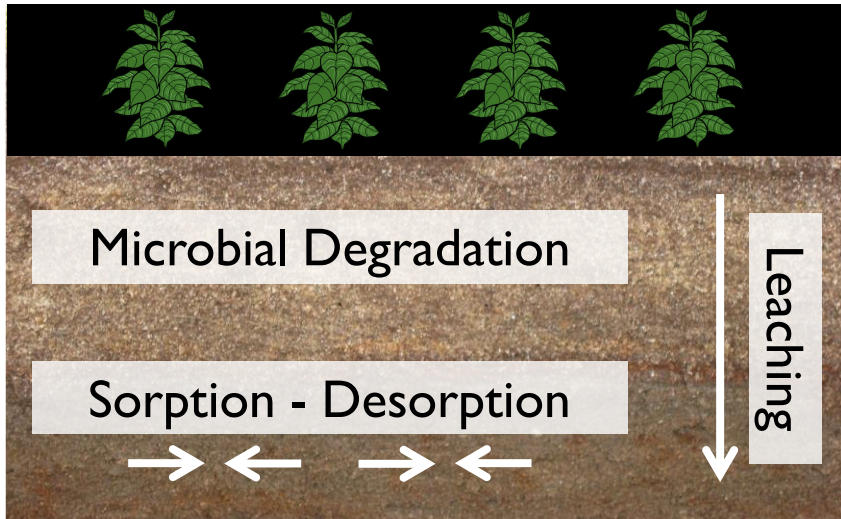
Volatilization

Plant Uptake & Metabolism

Microbial Degradation

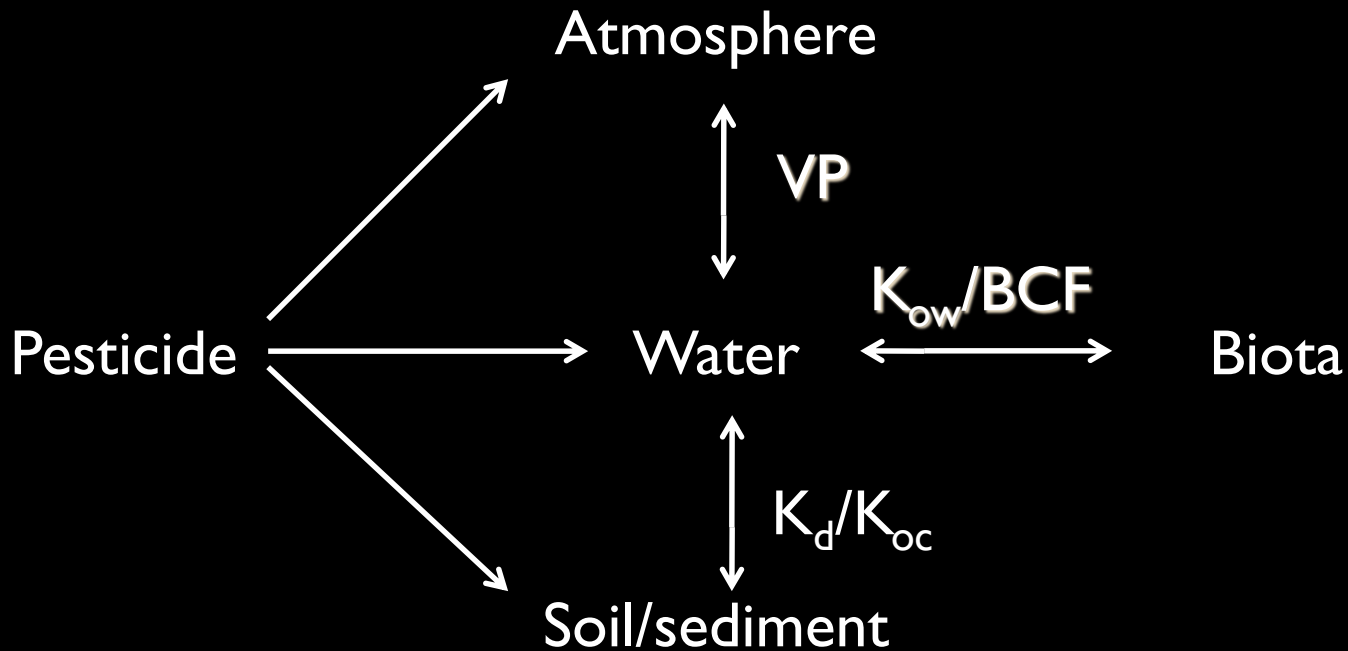
Sorption - Desorption

Leaching

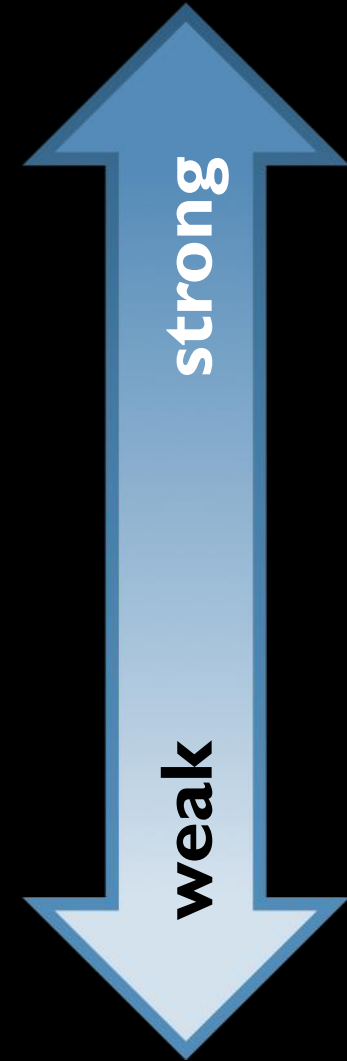
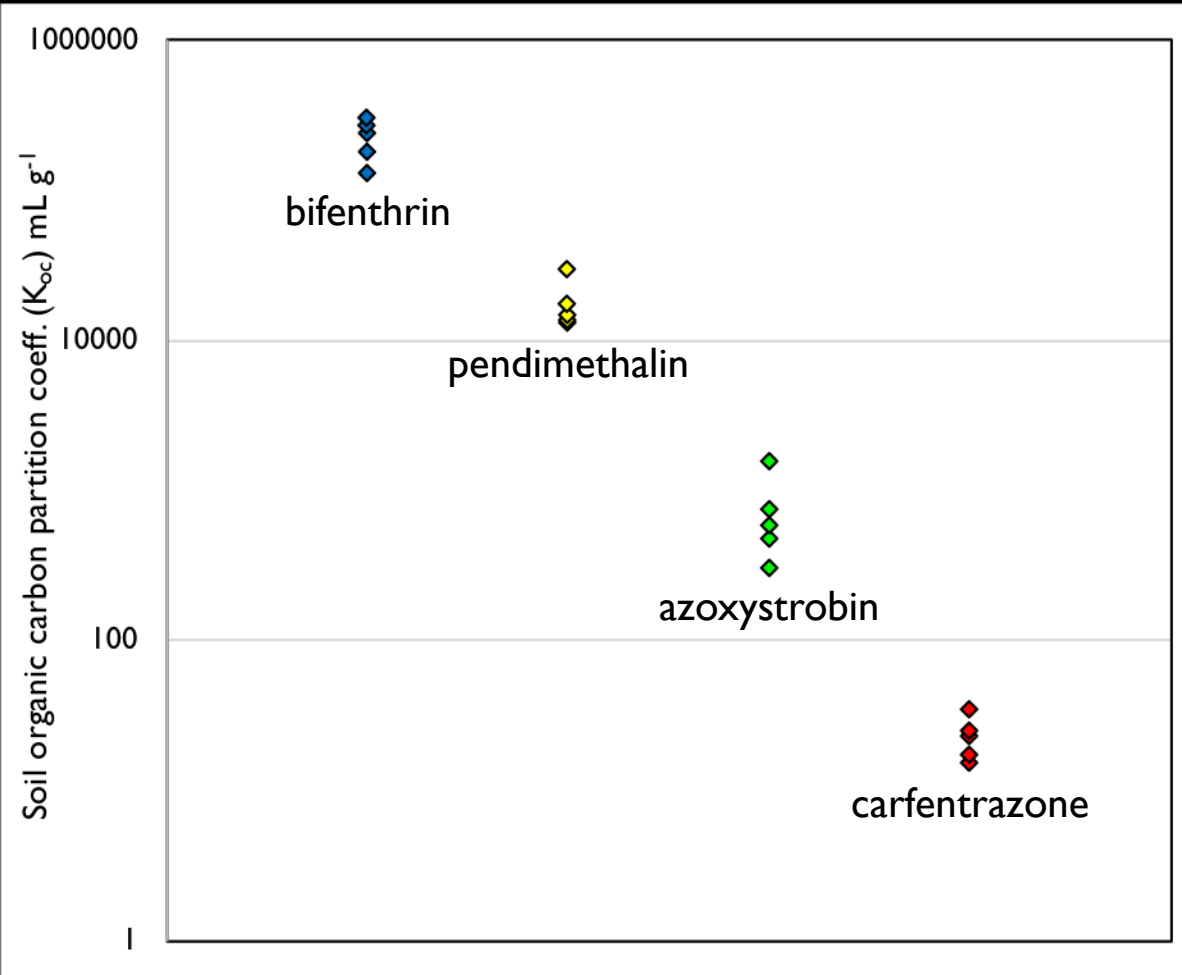




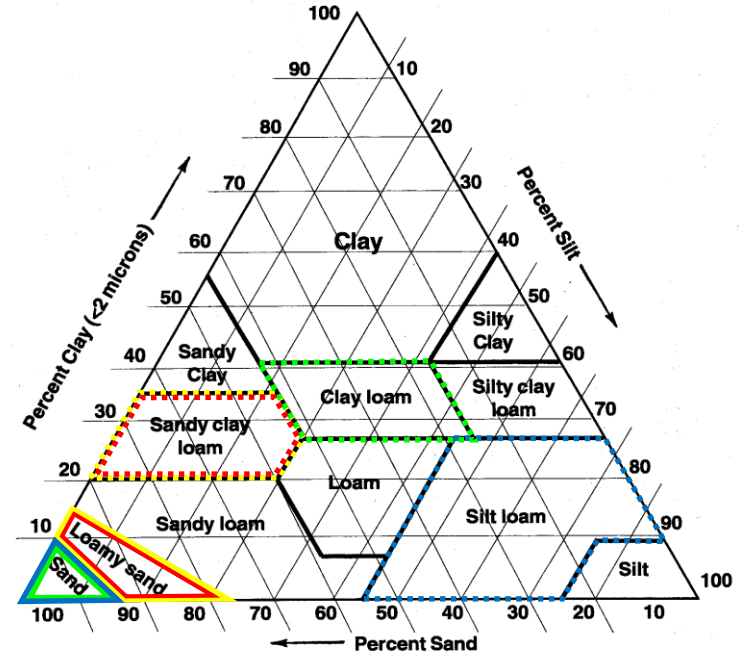
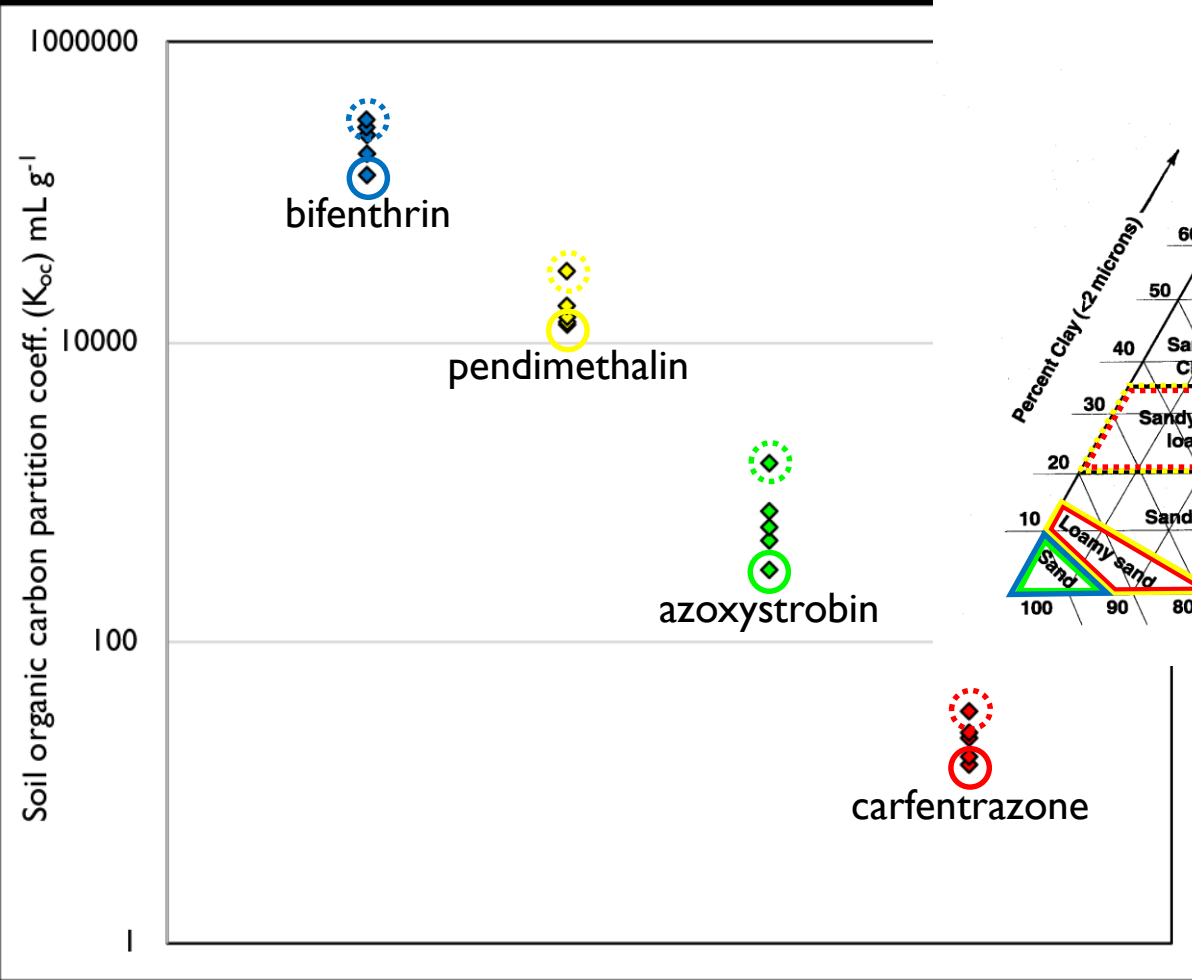
# What Happens After Pesticide Application



# Soil Adsorption



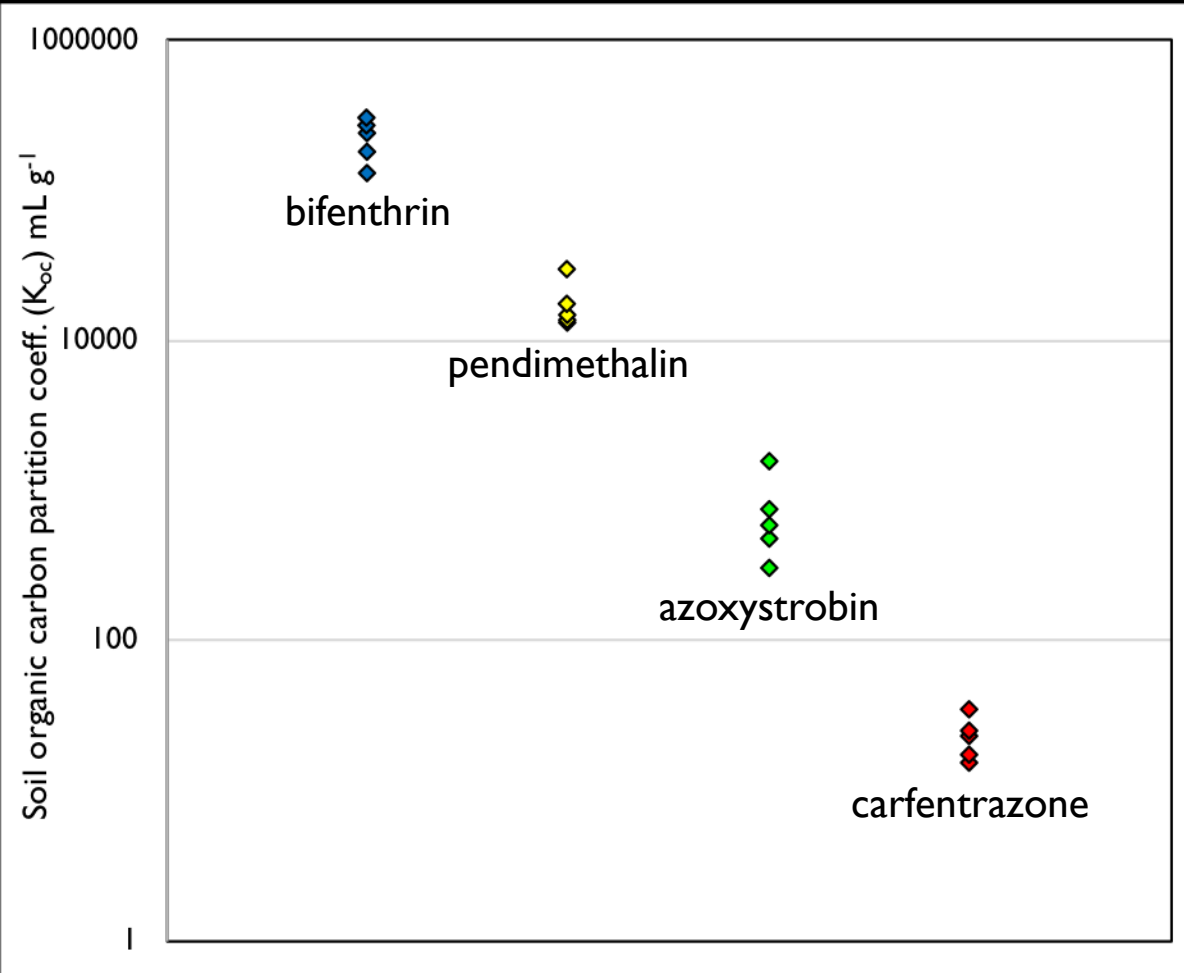
# Soil Adsorption



$K_{oc}$  values will vary depending on soil texture evaluated

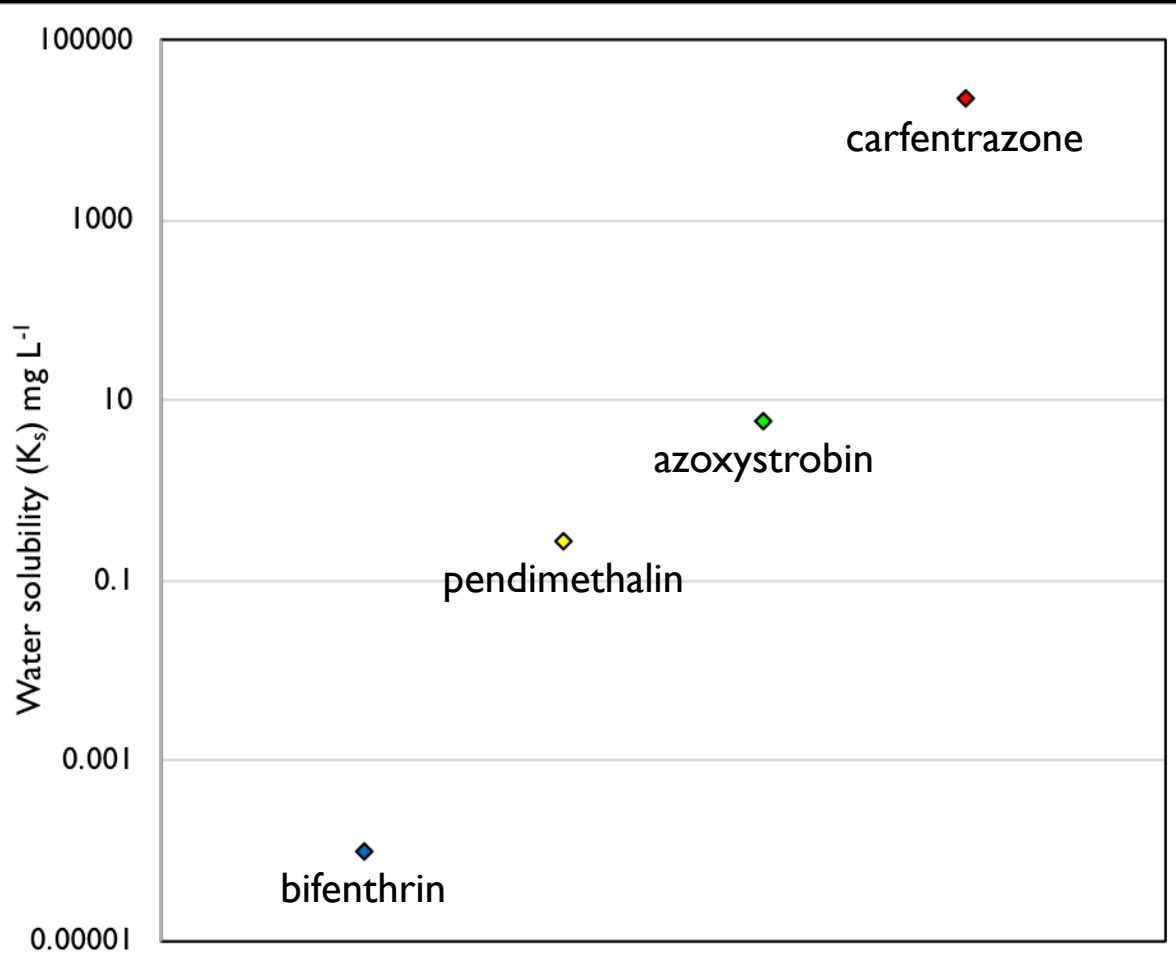


# Soil Adsorption



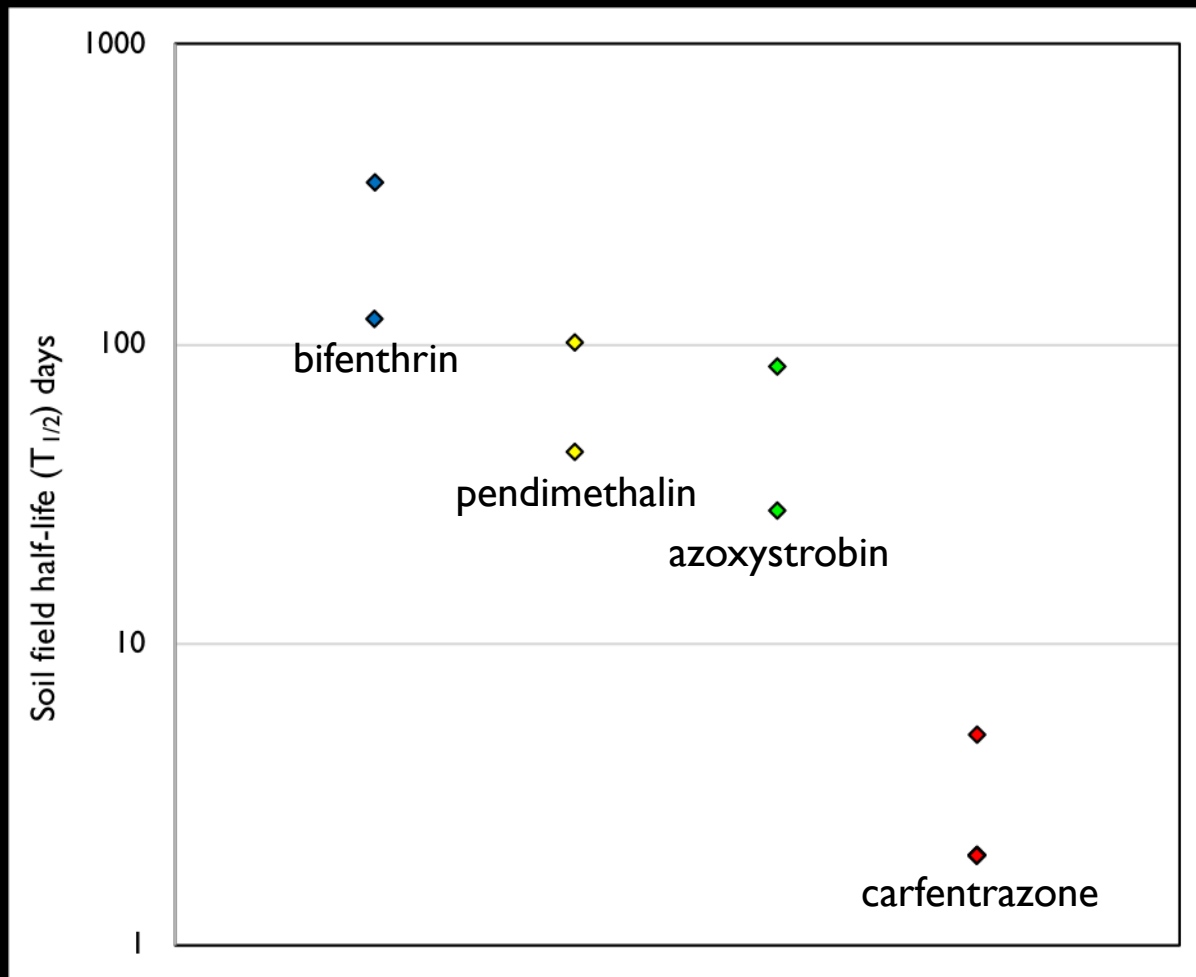
$K_{oc}$ (mL/g)	Solubility class
>5000	Very Strong
2000 - 5000	Strong
500 - 2000	Moderate
50 - 500	Weak
<50	Very Weak

# Water Solubility



$K_s$ (mg/L)	Solubility class
>5000	Very High
1000 - 5000	High
100 - 1000	Moderate
10 - 100	Low
<10	Very Low

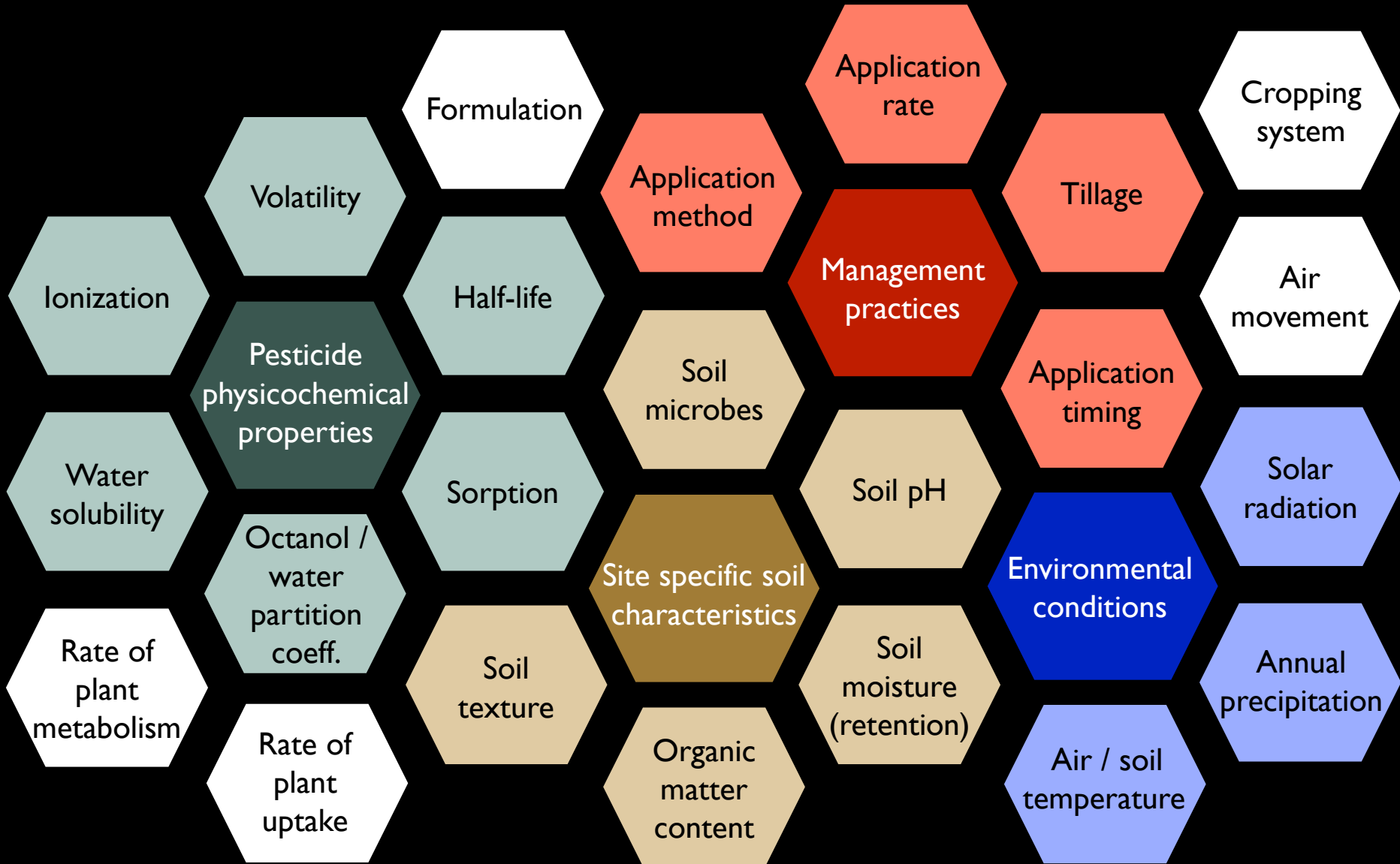
# Persistence



$T_{1/2}$ (days)	Persistence class
>150	Very Long
100 - 150	Long
30 - 100	Moderate
5 - 30	Short
<5	Very Short



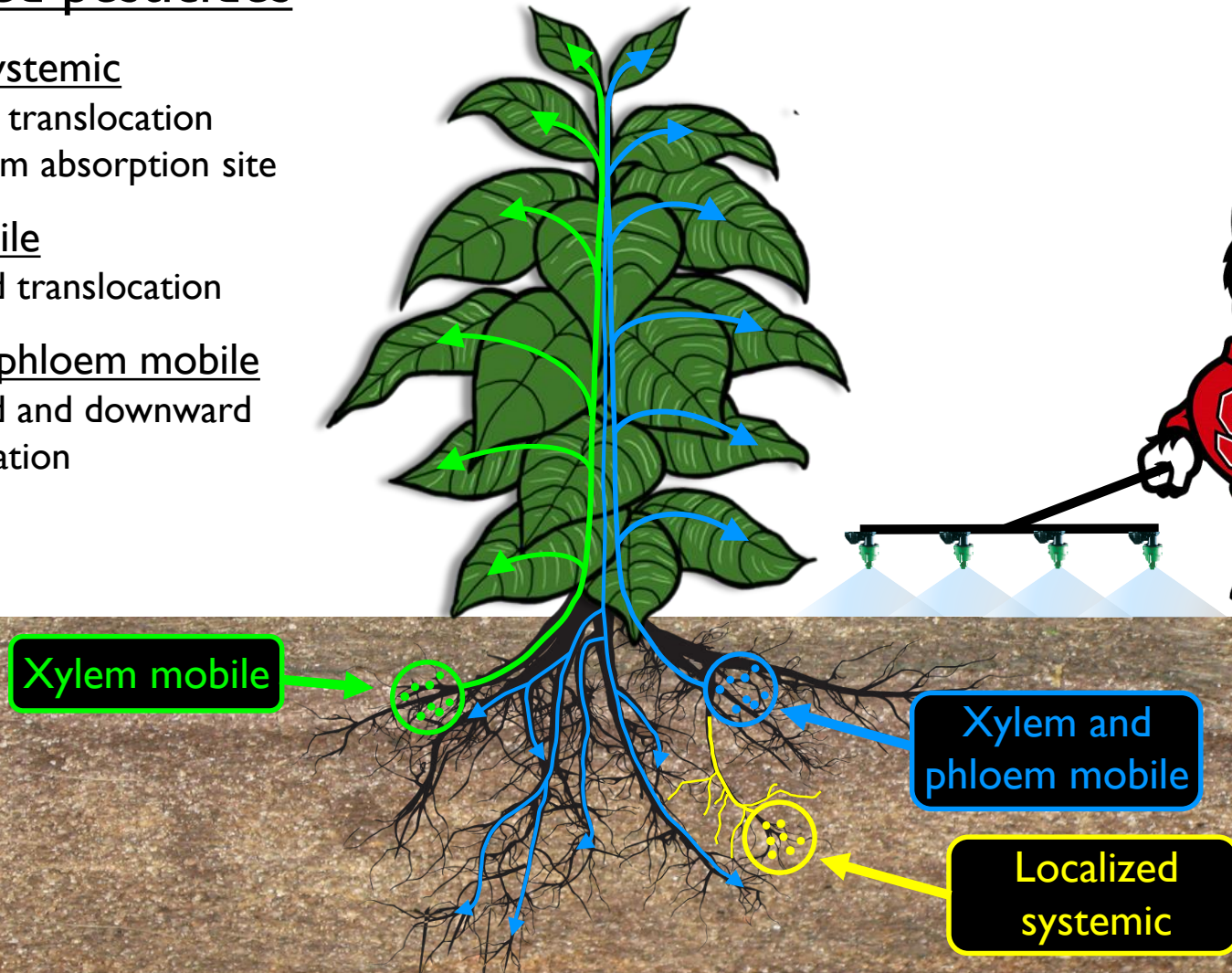
# Factors Affecting Pesticide Fate



# Root Uptake and Translocation

## Soil applied pesticides

- Localized systemic
  - Limited translocation from absorption site
- Xylem mobile
  - Upward translocation
- Xylem and phloem mobile
  - Upward and downward translocation



# Pesticide $K_{ow}$

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$$K_{ow} = \frac{\text{concentration in octanol phase (lipophilic)}}{\text{concentration in aqueous phase (lipophobic)}}$$

Octanol and water are immiscible, therefore this measurement serves as a relatively good approximation of the partitioning between the cytosol and lipid membranes of living systems.



# Pesticide $K_{ow}$

$K_{ow}$  values can be used to predict pesticide:

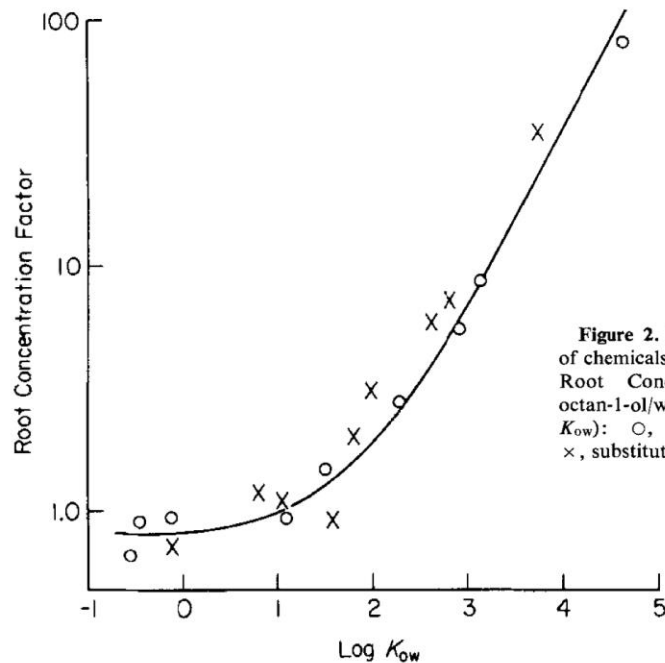
- water solubility
- sorption affinity
- propensity to bioaccumulate
- plant uptake and subsequent translocation

“ $K_{ow}$ ” is often presented as “ $\log K_{ow}$ ” with values ranging from -3 to 7

Pesticide	Log $K_{ow}$	Water solubility (mg/L)	$K_d$ Soil sorption (mL/g)
Acephate	- 0.9	790,000	0.9
Imidacloprid	0.6	610	132 to 310
Clomazone	2.5	1,110	150 to 562
Pendimethlin	5.3	0.275	17,200
Bifenthrin	6.0	0.001	131,000 to 302,000

# Effect of $\log K_{ow}$ on Absorption

Briggs et. al. 1982 investigated the relationship between lipophilicity and root uptake and translocation of nonionizable pesticides in barley.



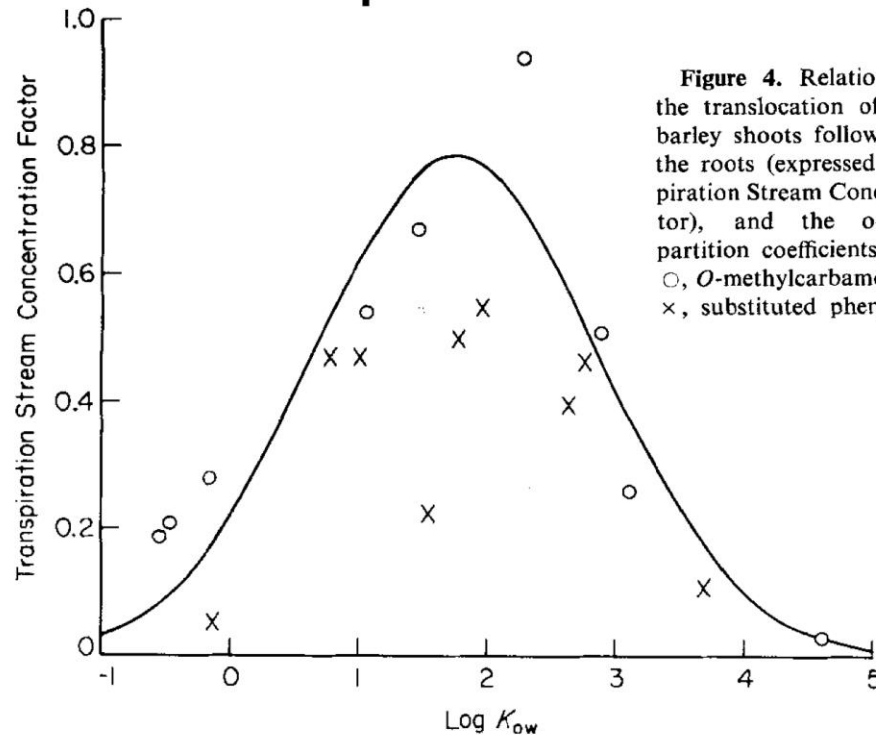
**Figure 2.** Relationship between the uptake of chemicals by barley roots (expressed as the Root Concentration Factor), and their octan-1-ol/water partition coefficients (as  $\log K_{ow}$ ):  $\circ$ , *O*-methylcarbamoyloximes; and  $\times$ , substituted phenylureas.



Root uptake of lipophilic chemicals is dominated by physical sorption.

# Effect of $\log K_{ow}$ on Translocation

Briggs et. al. 1982 investigated the relationship between lipophilicity and root uptake and translocation of nonionizable pesticides in barley.



**Figure 4.** Relationship between the translocation of chemicals to barley shoots following uptake by the roots (expressed as the Transpiration Stream Concentration Factor), and the octan-1-ol/water partition coefficients (as  $\log K_{ow}$ ):  $\circ$ , *O*-methylcarbamoyloximes; and  $\times$ , substituted phenylureas.

Transpiration stream concentration factor is estimated by determining the amount of the compound accumulating in the shoot per volume of water translocated.



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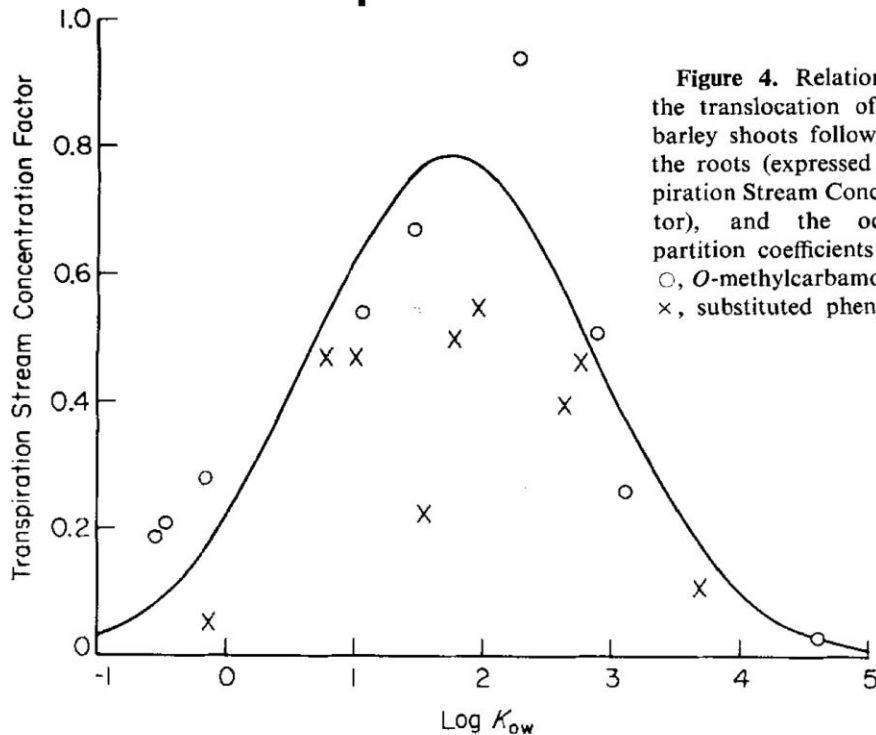
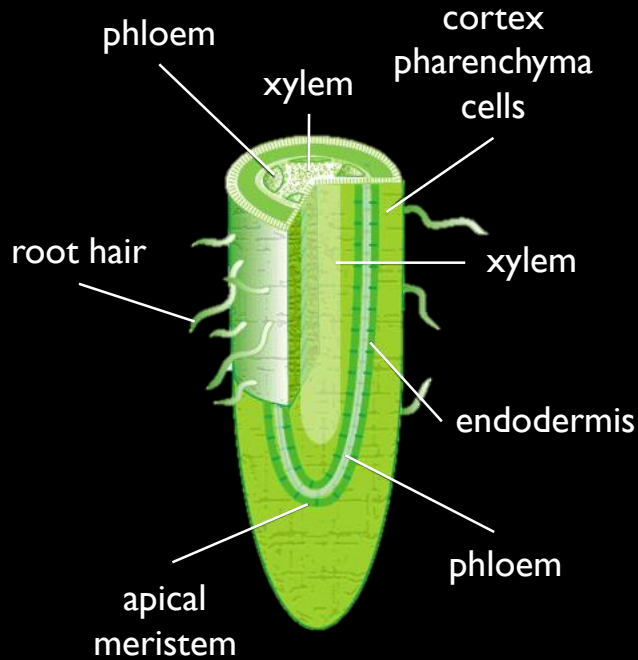


Figure 4. Relationship between the translocation of chemicals to barley shoots following uptake by the roots (expressed as the Transpiration Stream Concentration Factor), and the octan-1-ol/water partition coefficients (as  $\log K_{ow}$ ):  $\circ$ , *O*-methylcarbamoyloximes; and  $\times$ , substituted phenylureas.



Optimum  $\log K_{ow}$  for translocation is -0.5 to 3.5 (hydroponic culture).

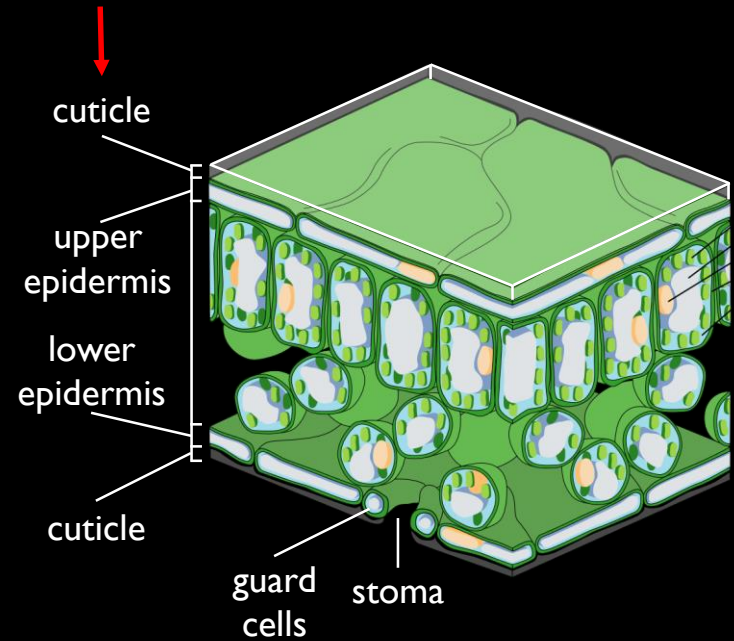
# Root vs. Leaf Morphology



## Root

- Absorb and translocate water and minerals to stem
- Store food

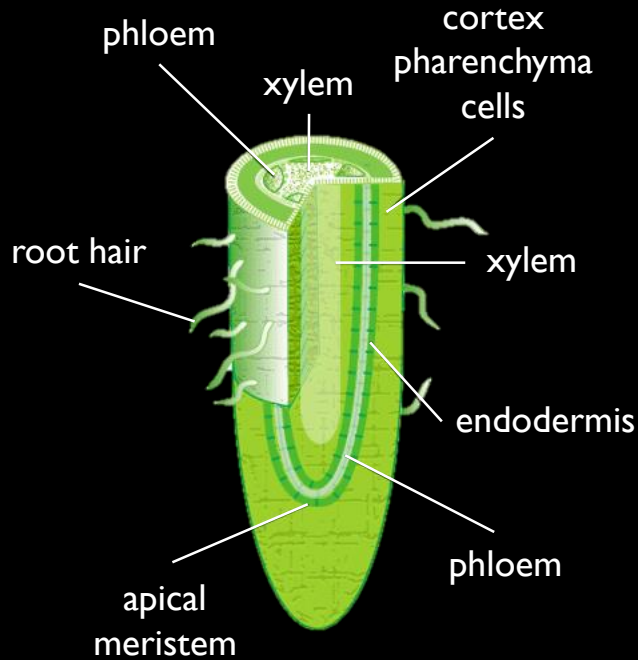
Embedded with waxes (give plants glossy appearance)



## Leaf

- Primary photosynthetic organ
- Prevent water loss
- Allow gas exchange

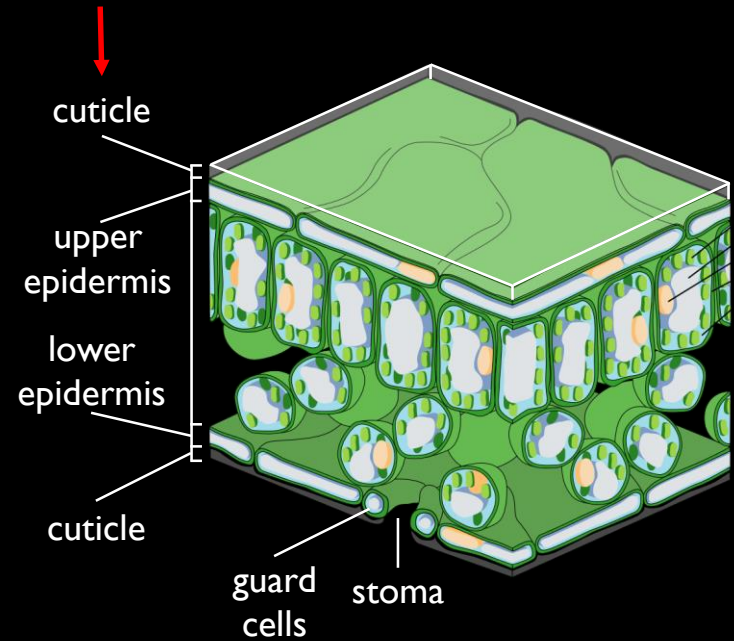
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# Foliar Absorption and Translocation

## Foliar applied pesticides

### • Contact

- No translocation

### • Localized systemic

- Limited translocation from absorption site

### • Xylem mobile

- Upward translocation

### • Xylem and phloem mobile

- Upward and downward translocation





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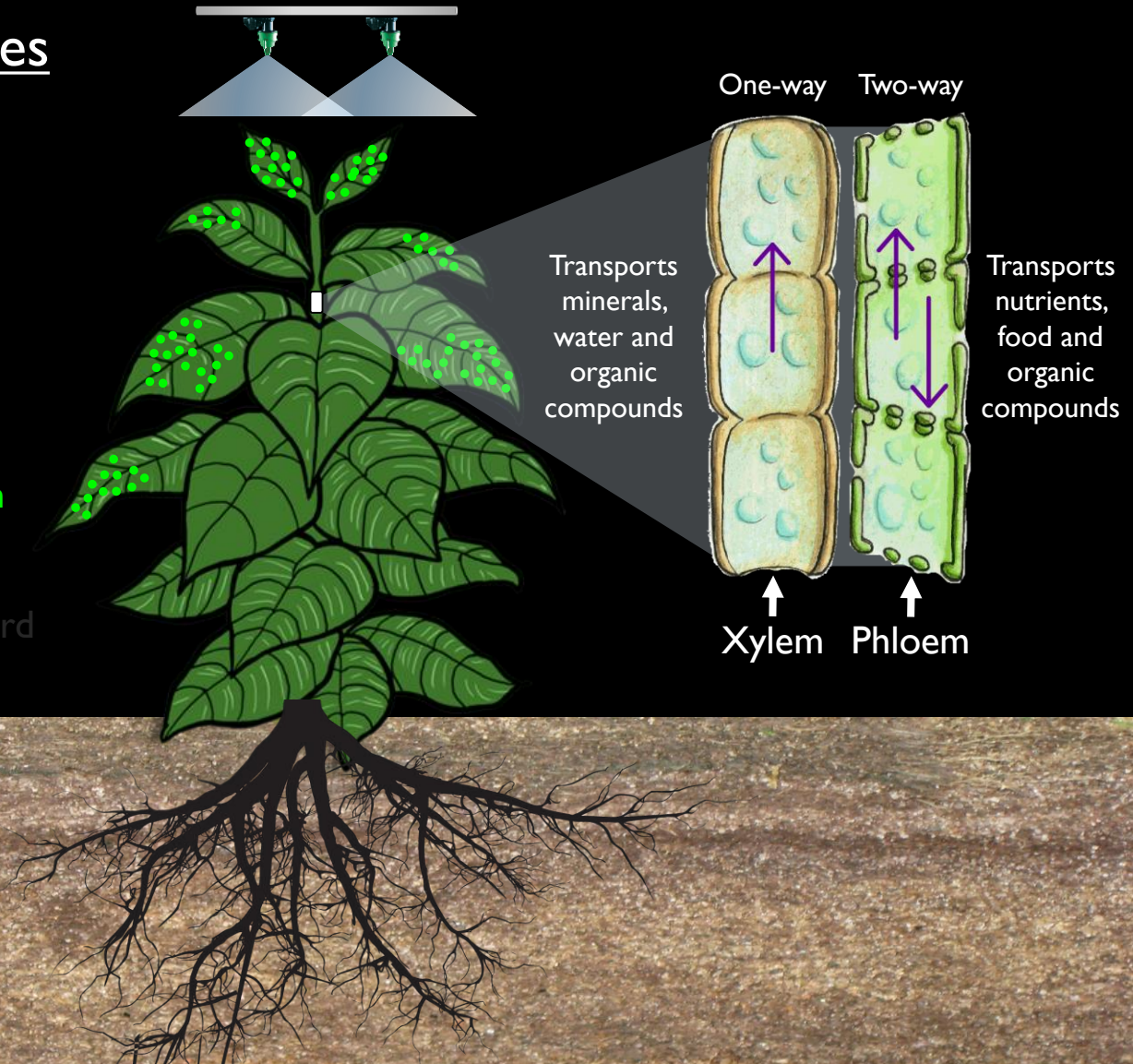
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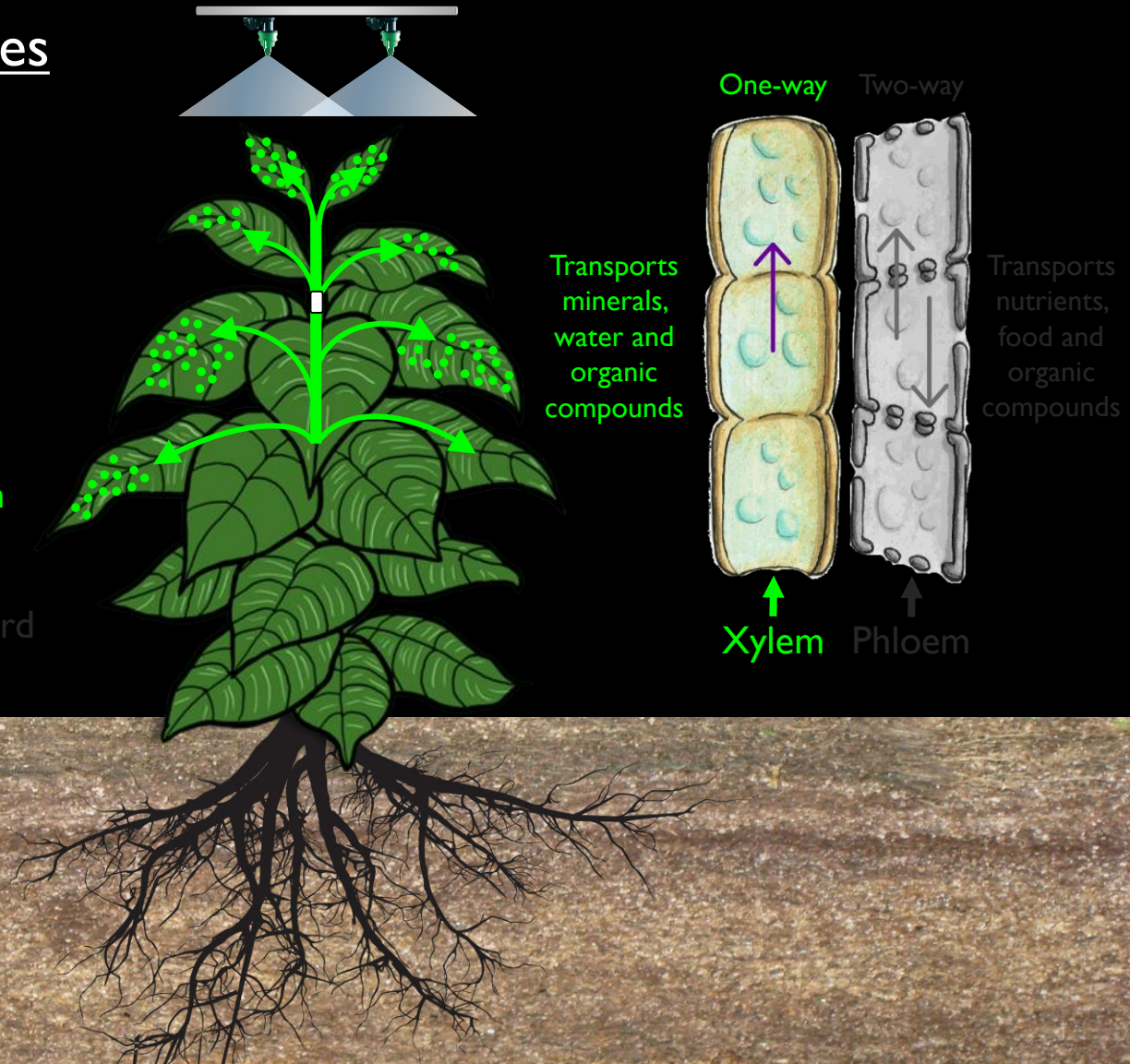




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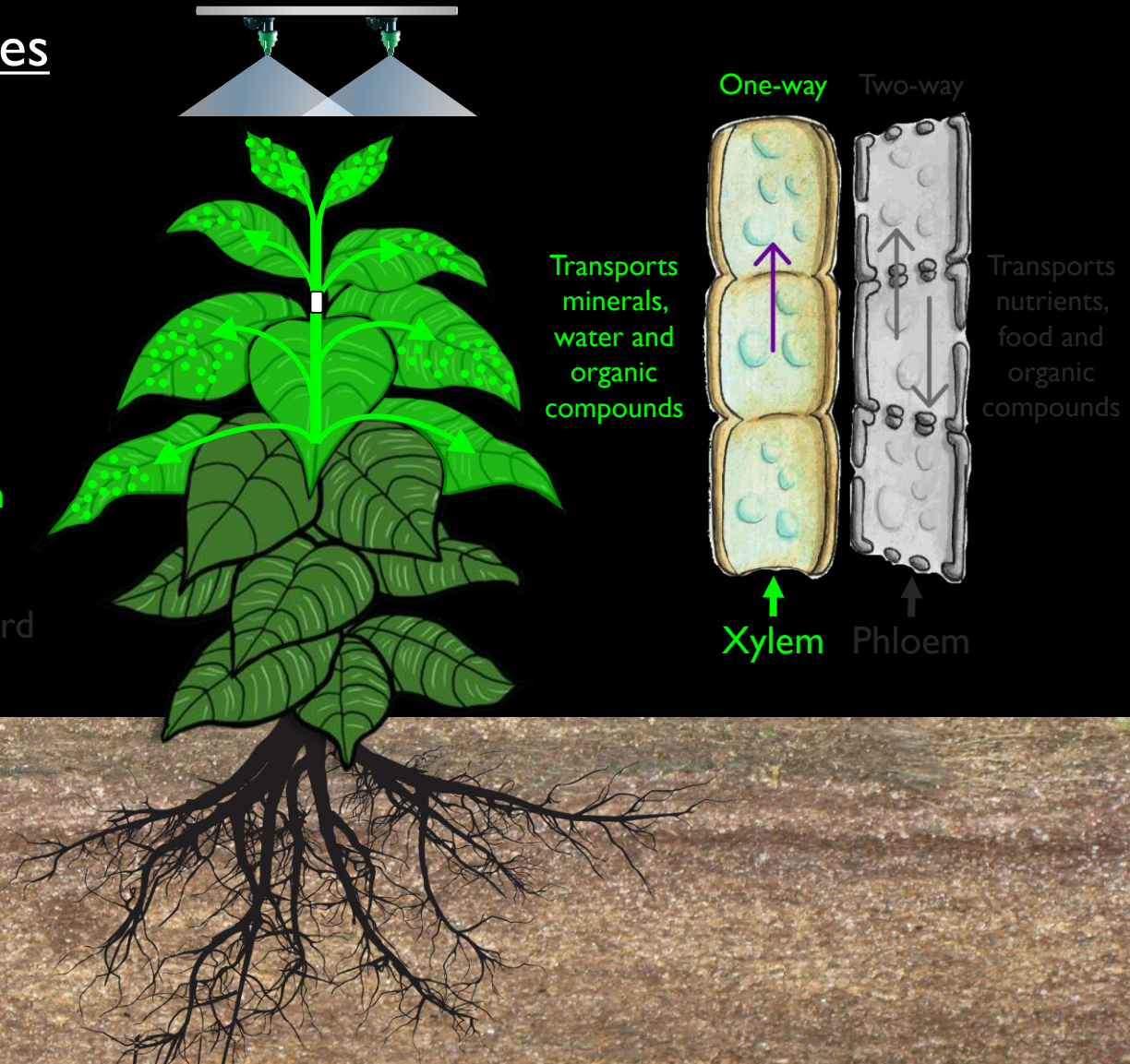




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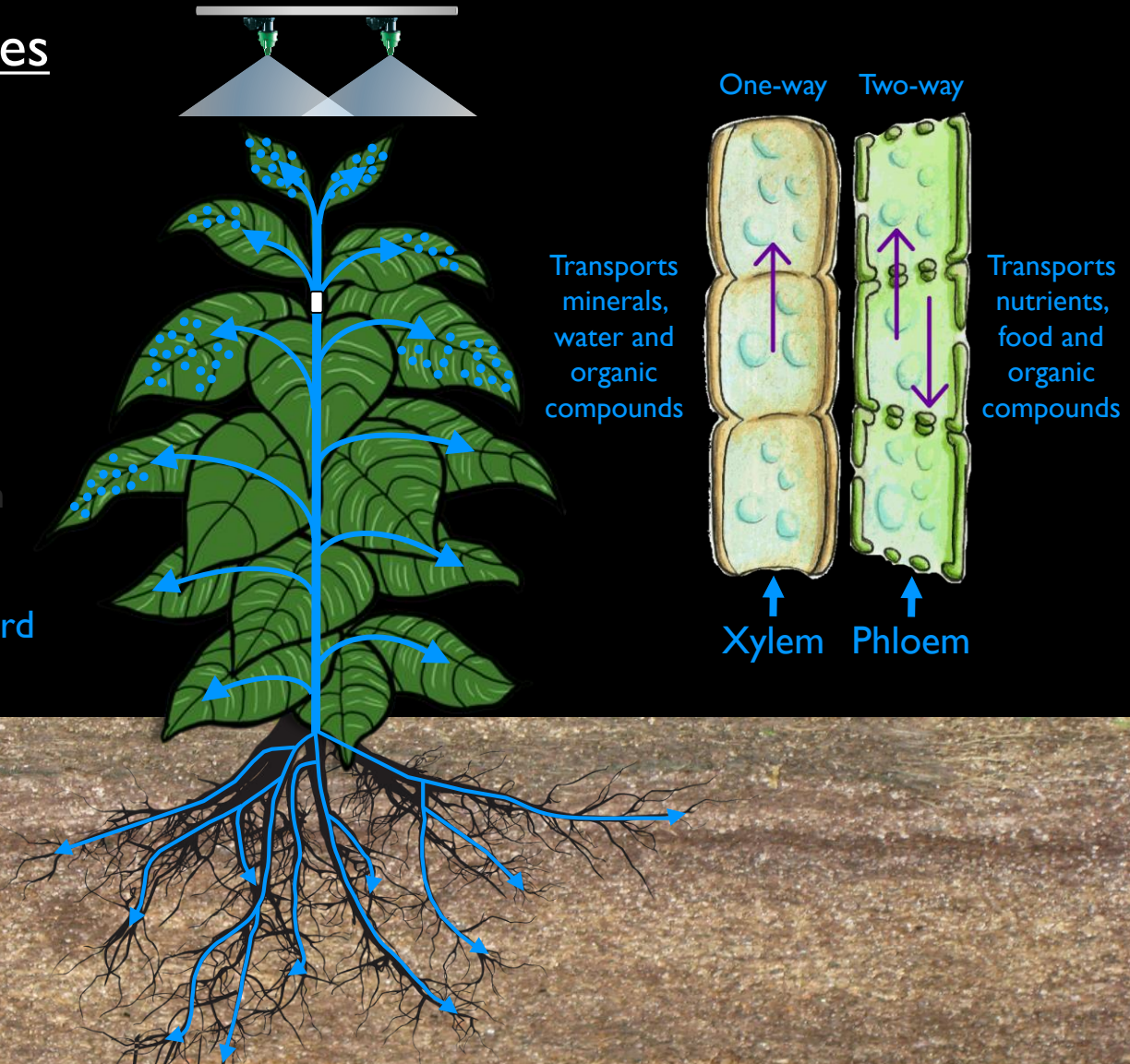




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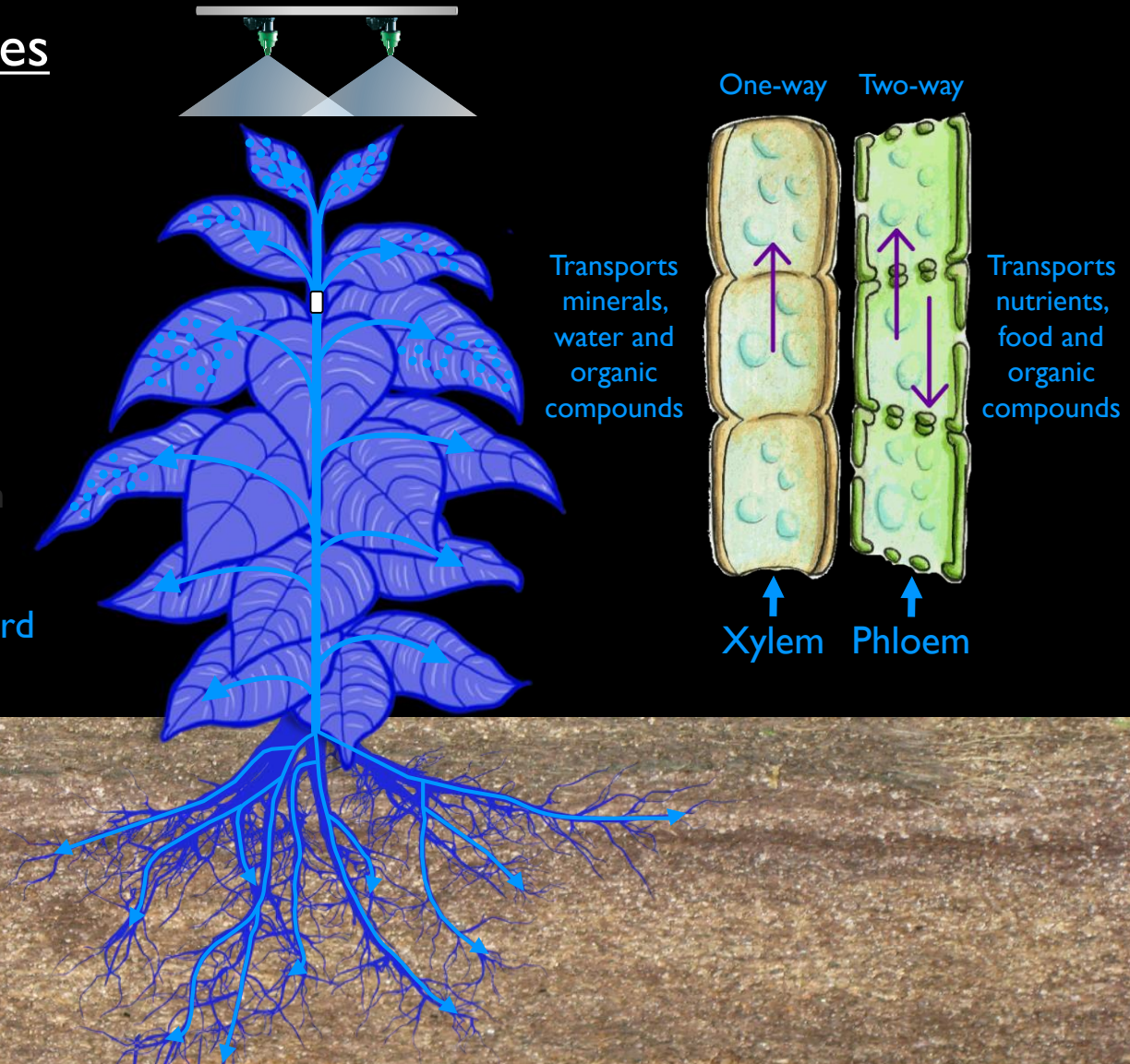




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# Pesticide Persistence in Foliage Affected by...

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- Many factors including but not limited to:
  - Physicochemical properties of pesticide
    - Volatility, solubility, etc.
  - Environmental factors
    - Temperature, precipitation, humidity, air movement, etc.
  - Application timing
    - Rate of growth



# Herbicide Properties

Herbicide	Soil sorption $K_{oc}$ (mL/g)	Persistence soil field $T_{1/2}$ (days)	Translocated via:	Water solubility $K_s$ (mg/L)
Carfentrazone	15 to 25	2.4 hours	Limited	12,000
Clomazone	150 to 562	5 to 60	Limited	1,100
Maleic hydrazide	23 to 250	2 to 7	Xylem and phloem	300,000
Napropamide	600	70	Xylem	73
Pebulate	430	36 to 60	Xylem	60
Pendimethalin	17,200	44	Limited	0.275
Sethoxydim	100	5	Xylem and phloem	1,040
Sulfentrazone	43	121 to 302	Limited	110 to 1,600

# Insecticide Properties

Insecticide	Soil sorption $K_{oc}$ (mL/g)	Persistence soil field $T_{1/2}$ (days)	Translocated via:	Water solubility $K_s$ (mg/L)
Acephate	0.9	2 to 7	Limited	790,000
Acetamiprid	71 to 313	8.2	Xylem and phloem	2,950
Azadirachtin	21 to 875	2 to 25	Xylem	260
Bifenthrin	236,750	122 to 345	Limited	0.001
Chlorantraniliprole	328,000	1130	Xylem and phloem	1.0
Cyantraniliprole	128 to 266	32.3	Xylem and phloem	14.2
Imidacloprid	132 to 310	27 to 229	Xylem and phloem	610
Lambda-cyhalothrin	180,000	28 to 84	Xylem	0.0042
Pymetrozine	2,245	14	Xylem and phloem	290
Spinosad	4,227	9 to 17	Localized systemic	225

# Fungicide Properties

Fungicide	Soil sorption $K_{oc}$ (mL/g)	Persistence soil field $T_{1/2}$ (days)	Translocated via:	Water Solubility $K_s$ (mg/L)
Acibenzolar	1,394	20	Xylem and phloem	7.7
Azoxystrobin	210 to 580	28 to 85	Xylem	6.7
Dimethomorph	290 to 566	35 to 53	Xylem (localized)	49.2
Fluopicolide	25,000	107	Xylem	2.8
Fosetyl-Al	0.1	1 hour	Xylem and phloem	111,300
Mancozeb	998	1 to 7	Limited	6.2
Mandipropamid	847	75 to 101	Localized systemic	4.2
Mefenoxam	20 to 790	42	Xylem	26,000

# Pesticide Properties - Review

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Imidacloprid	132 to 310	27 to 229	Xylem and phloem	610
Lambda-cyhalothrin	180,000	28 to 84	Xylem	0.0042
Clomazone	150 to 562	5 to 60	Limited	1,100
Pendimethalin	17,200	44	Limited	0.275
Sethoxydim	100	5	Xylem and phloem	1,040
Sulfentrazone	43	121 to 302	Limited	110 to 1,600 (pH)



# Summary

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- Pesticide fate and residue will become increasingly more important
- Pesticide fate depends on various factors
  - Pesticide physicochemical properties
  - Site specific soil characteristics
  - Management practices
  - Environmental conditions, among others
- Important to understand systemic nature of pesticides
  - Efficacy
  - Residue



Thank you!

NC State University  
Pesticide Environmental Fate Lab

*Travis Gannon*