

# Lina Cai

Factors affecting carbon filter  
adsorption in cigarette smoke



# **Factors Affecting Carbon Filter Adsorption in Cigarette Smoke**

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# Outline



➤ **Background**

➤ **Experimental design**

➤ **Results and discussion**

➤ **Conclusions**

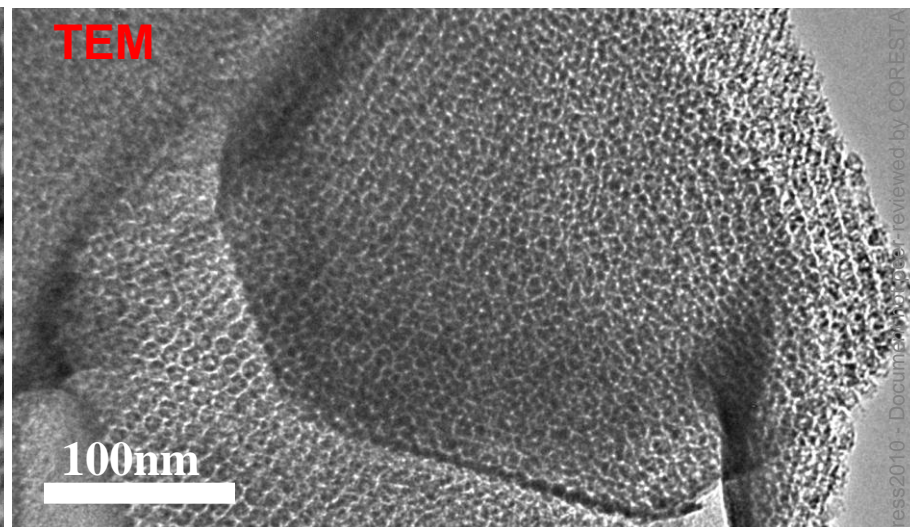
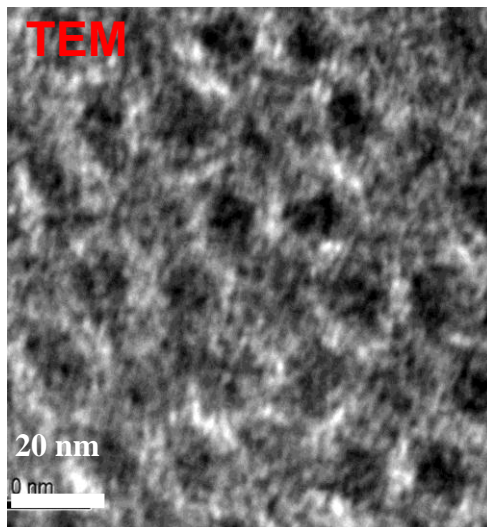
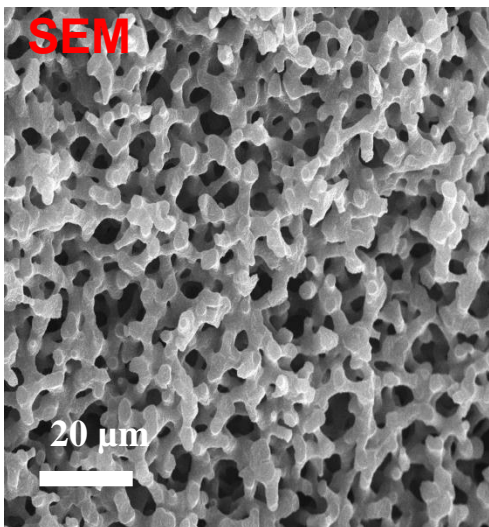
# Background

**Porous carbon can be used as a very efficient adsorbent in the cigarette filter**

**Because of**



- **Good absorbability**
- **Excellent thermal stability**
- **Variable surface chemistry**



# Research motivations

- **To investigate the contribution of tar deposition to decreasing carbon activity in a cigarette filter**
- **To establish the relative constructions towards porous carbon deactivation from the smoke particulate phase and the smoke vapor phase**
- **To find out which carbon parameters give the greatest adsorption in the cigarette filter and which types of compounds are adsorbed**

# Experimental design



**cigarette**

**self-made filter**

**pad**

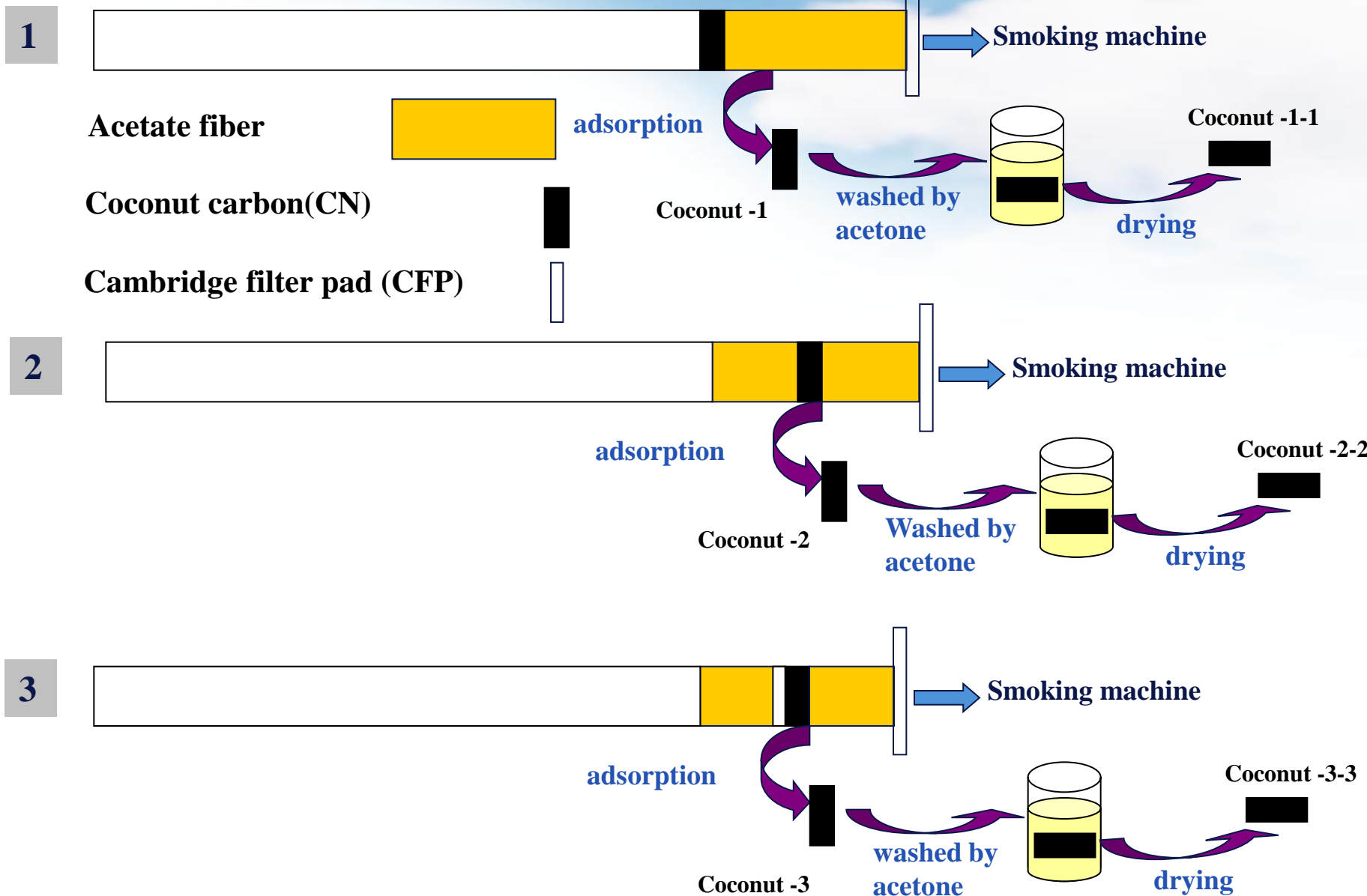


**gasbag**



**smoking engine**

# Experimental design



# Characterization



## ➤ Nitrogen sorption analysis

To get the pore structure information of the carbon



## ➤ Thermogravimetric analysis

To get the thermal information of the carbon



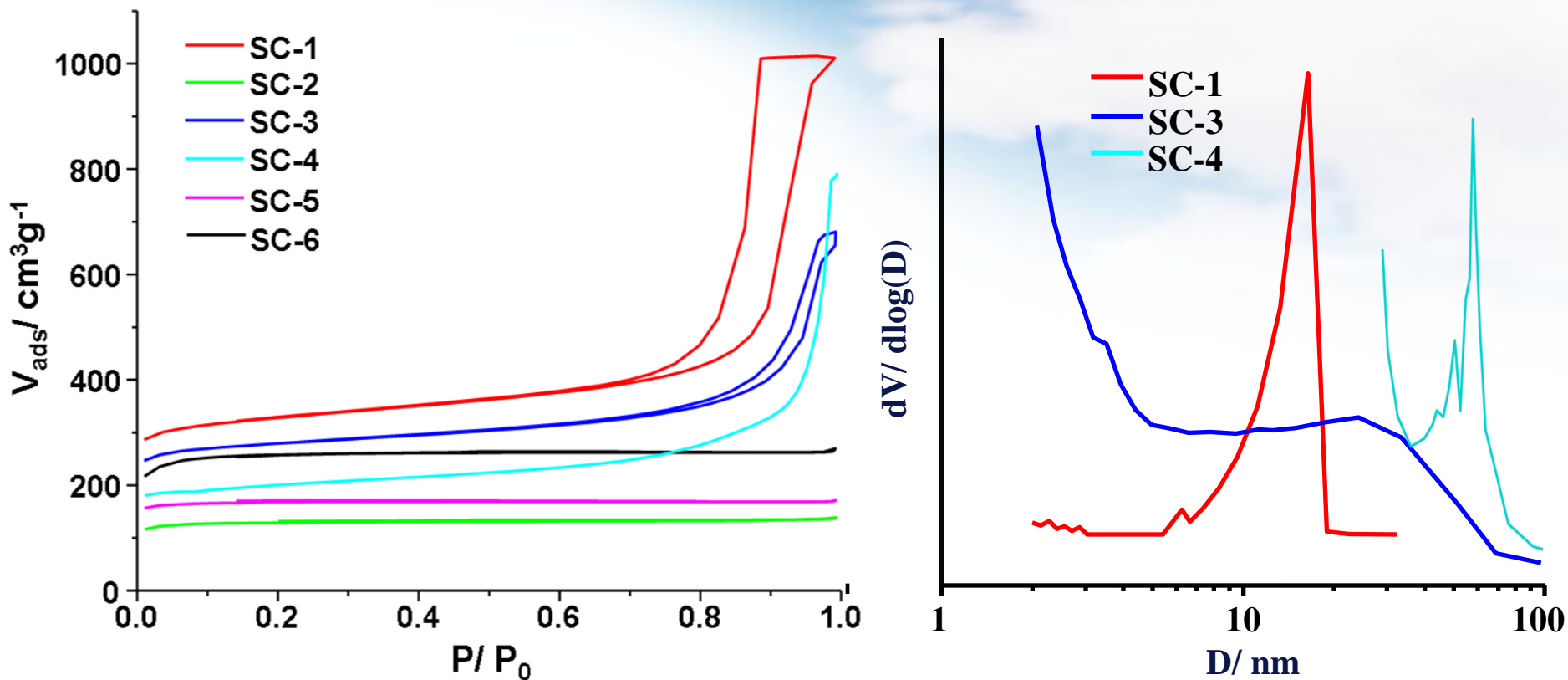
## ➤ Gas chromatography analysis

To get the absorbability information of the carbon





# Results and discussion



**Nitrogen sorption isotherms and pore size distributions of the porous carbons**

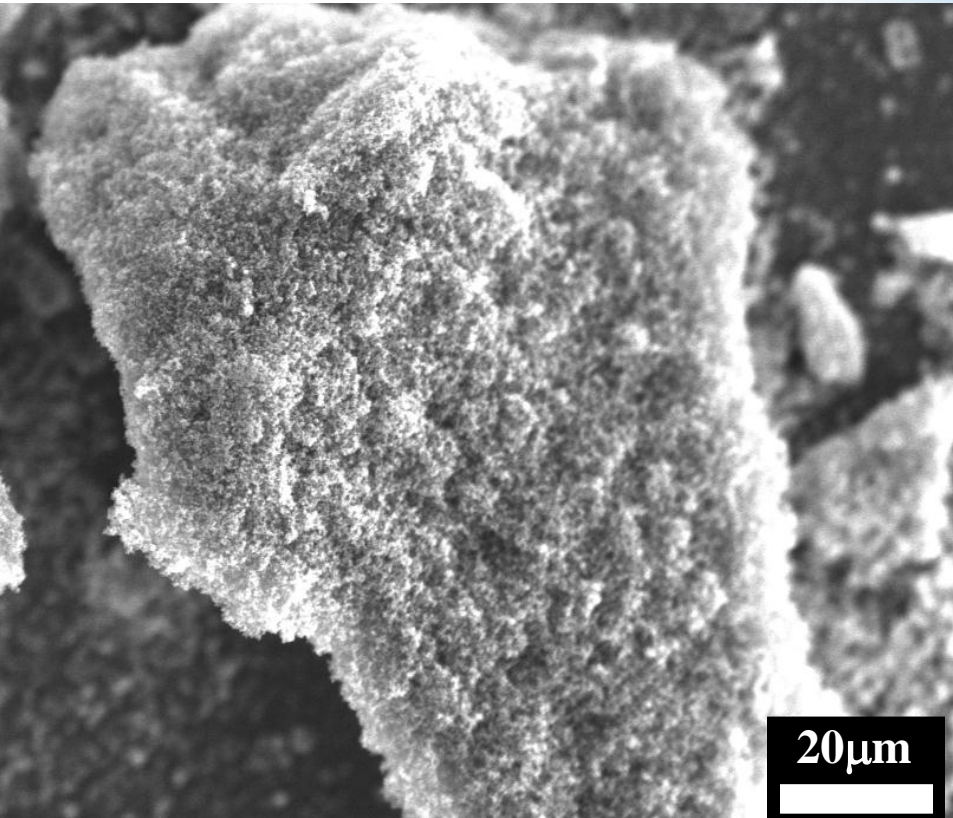
# Texture parameters of the carbons



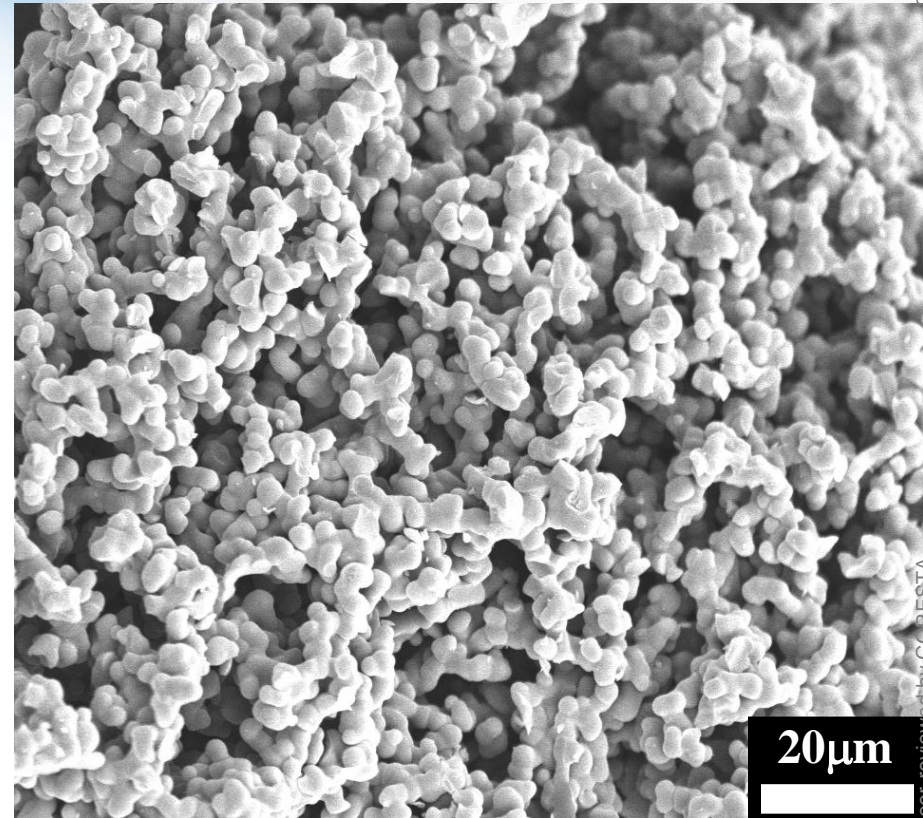
Sample	$S_{\text{BET}}$ ( $\text{m}^2/\text{g}$ )	$V_{\text{total}}$ ( $\text{cm}^3/\text{g}$ )	$V_{\text{mic}}$ ( $\text{cm}^3/\text{g}$ )	$D_{\text{peak}}$ (nm)
SC-1	590	1.12	0.16	13.2
SC-2	440	0.22	0.19	-
SC-3	610	0.81	0.18	23.8
SC-4	568	0.63	0.17	55
SC-5	463	0.23	0.22	-
SC-6(Coconut carbon)	867	0.40	0.34	-

Five different synthetic porous carbons and one coconut carbon were used as adsorbents. They had clearly different specific surface areas and pore size distributions.

# SEM observation



SC- 3



SC- 5

SEM images of porous carbons with different pore structure

# Position of the carbons in the filter



1

2

3

4

5

Filter

Acetate fiber

Coconut carbon

Cambridge filter pad

Cigarette

a

b

c

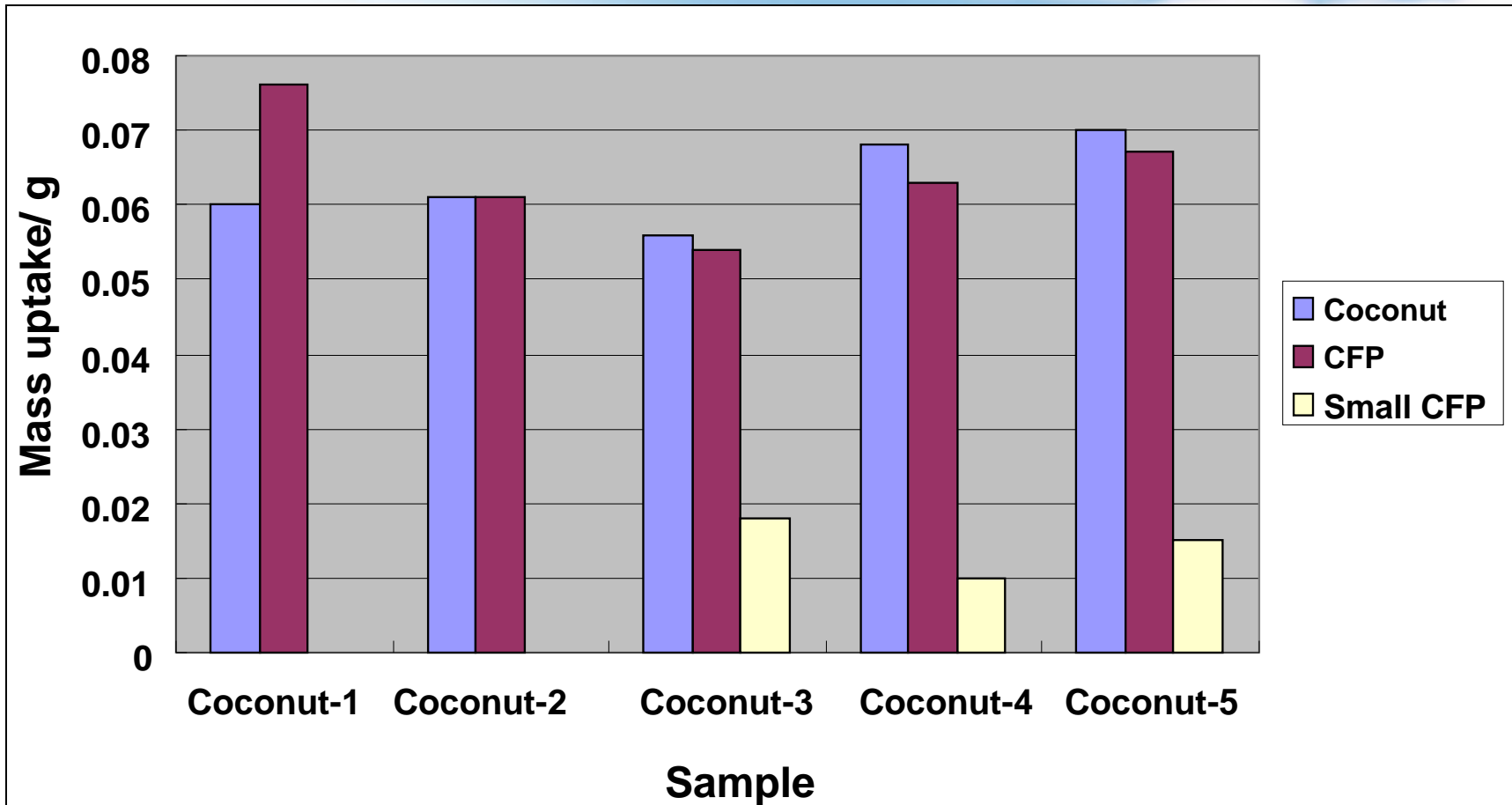
d

e

f

Schematic illustrations for the position of the carbons in the filter

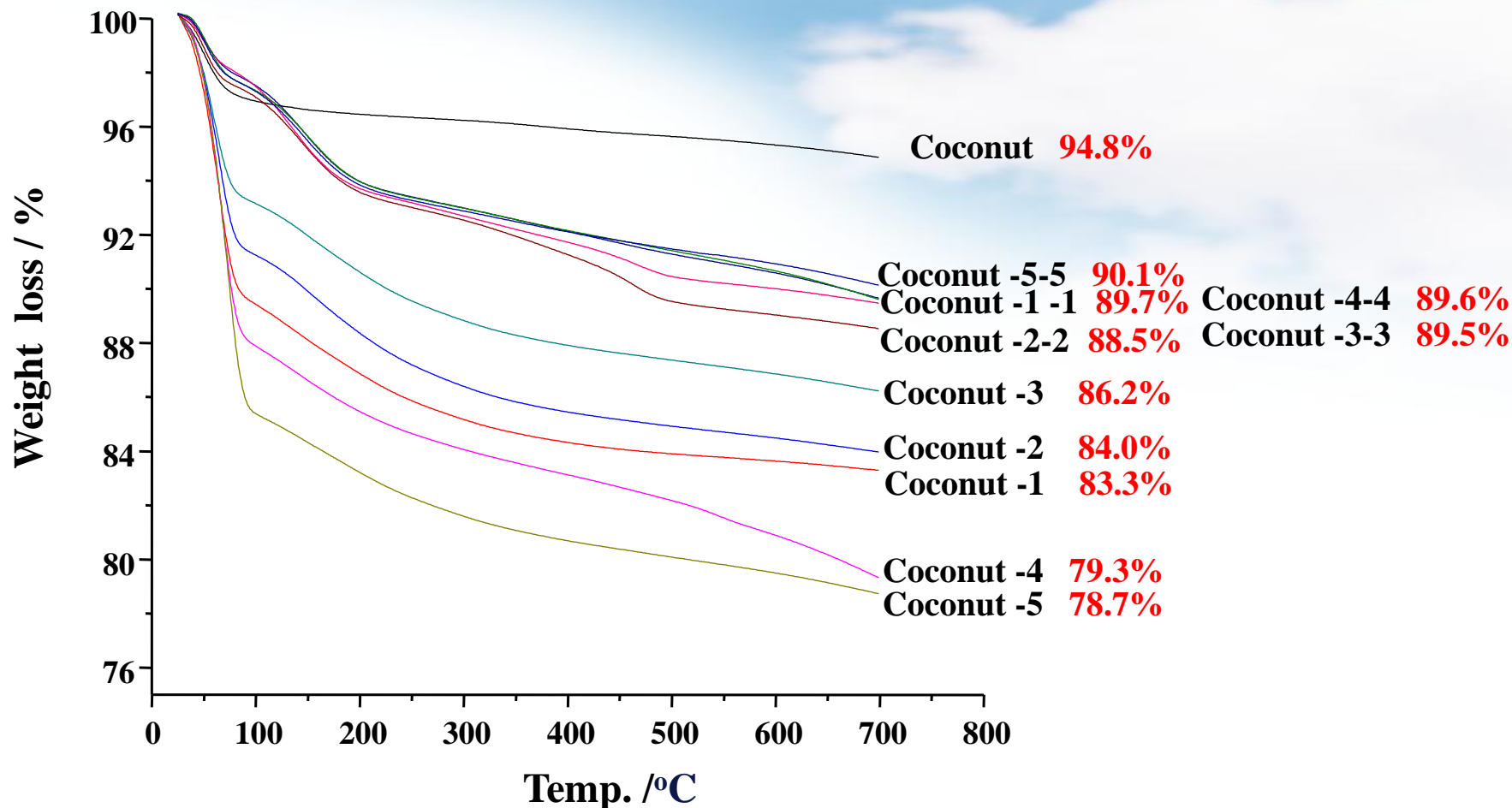
# Mass uptakes of the carbons and pads



**Mass uptakes for coconut carbons and Cambridge filter pads**

**Tar deposition on coconut carbon caused a negative impact on the carbon adsorption capacity.**

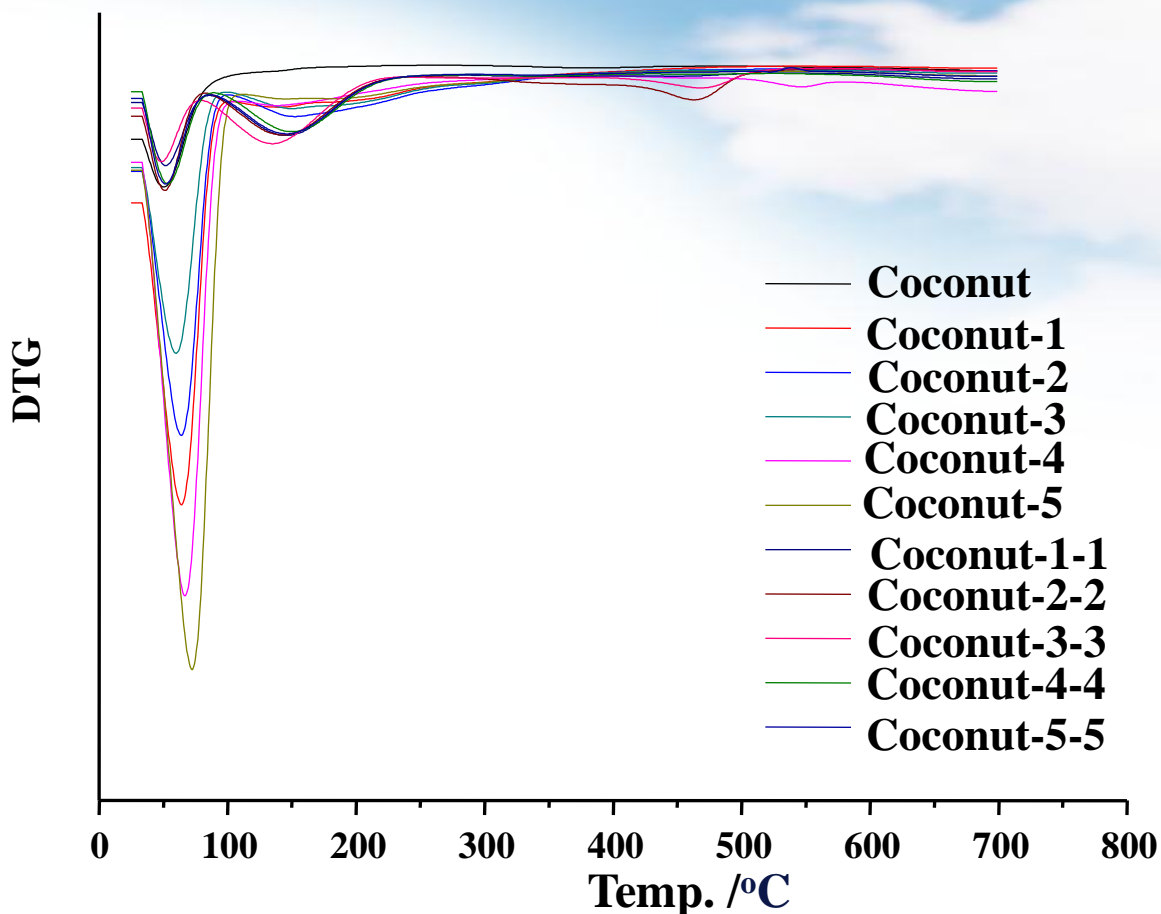
# TG curves



**TG curves of coconut after smoking**

More pieces of small Cambridge filter pads placed in front of carbon resulted in a greater weight loss of the carbon, indicating a better adsorption performance of the carbon.

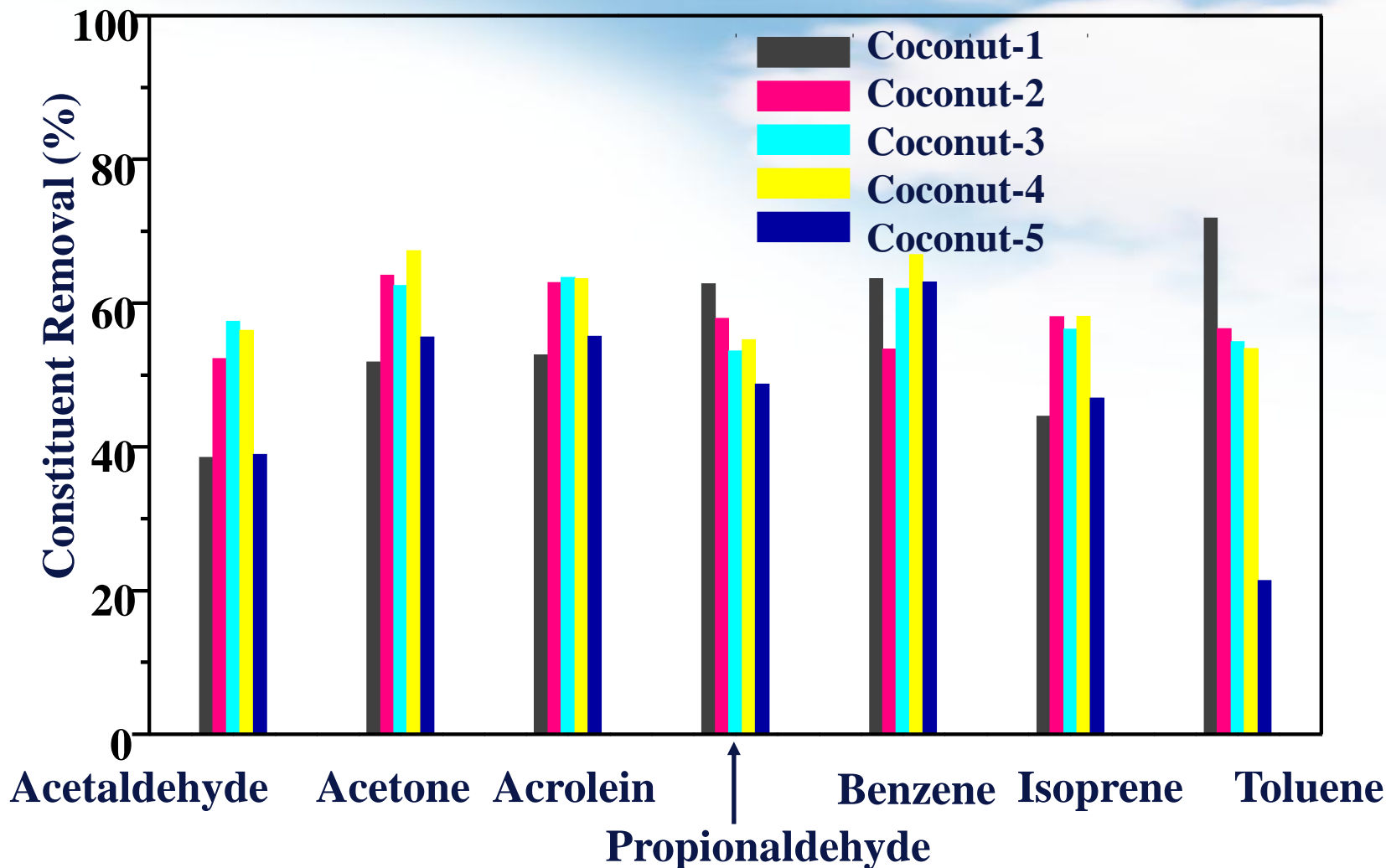
# DTG curves



## DTG curves of coconut after smoking

Samples showed a very fast mass loss at ca. 63 °C. Most of the adsorbed compounds volatilized from the coconut carbon around this temperature.

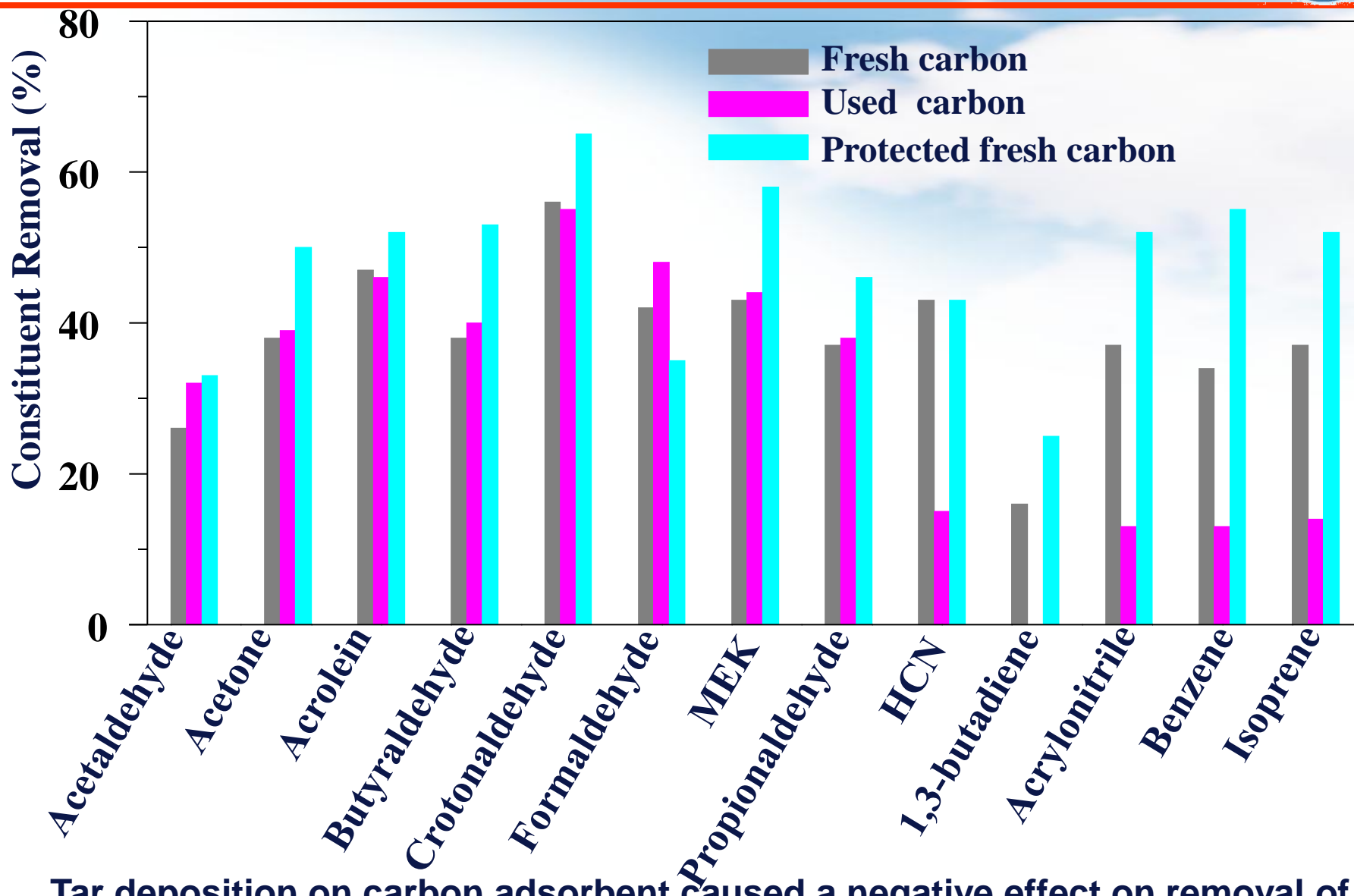
# Removal efficiency



The tar deposition on the surface of coconut carbon prevented further absorption of toxicants in the smoke vapor phase.

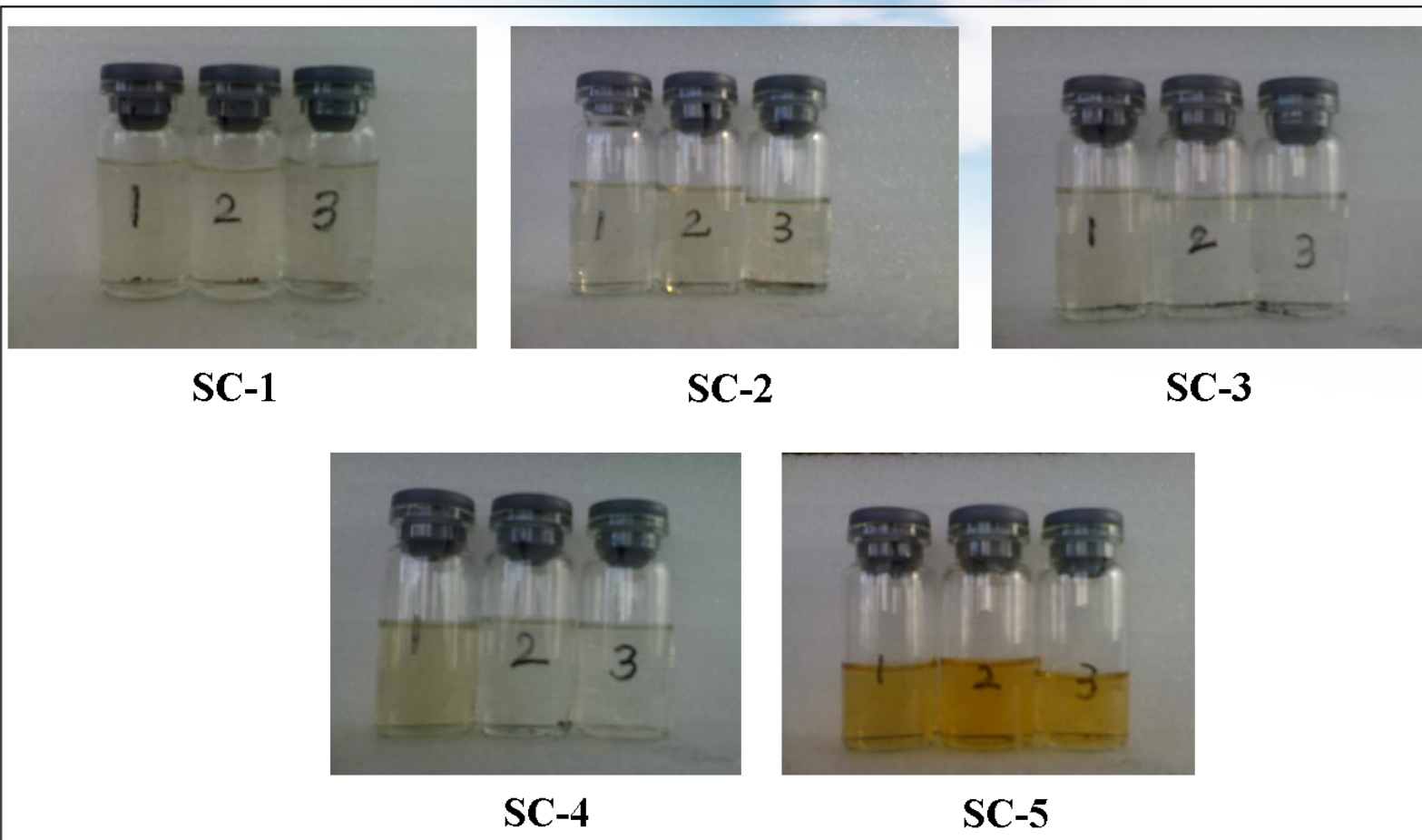


# Removal efficiency



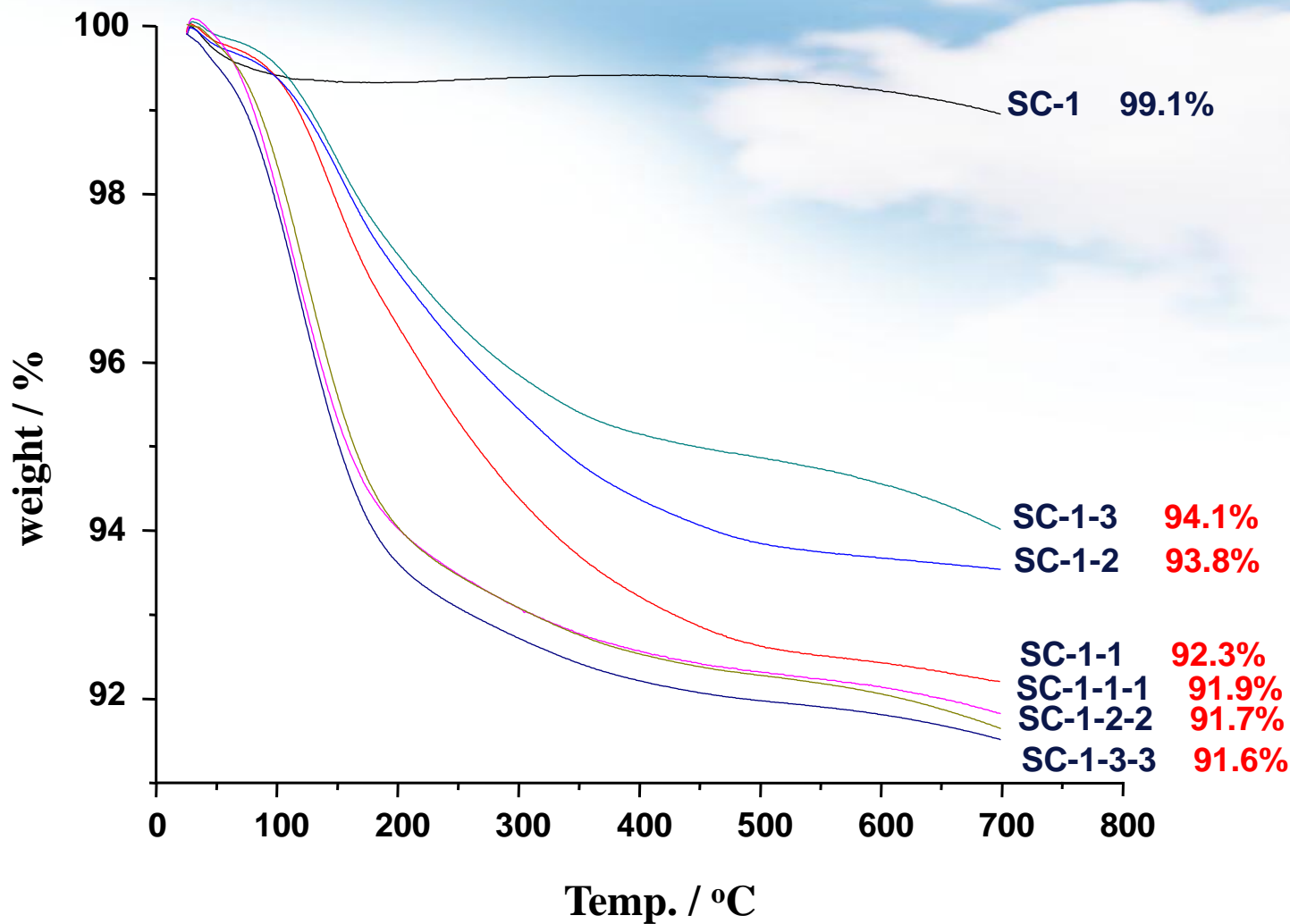
Tar deposition on carbon adsorbent caused a negative effect on removal of the majority toxicants in a vapor

# Carbon extraction by acetone



Photos of the extracted acetone solutions of different carbons

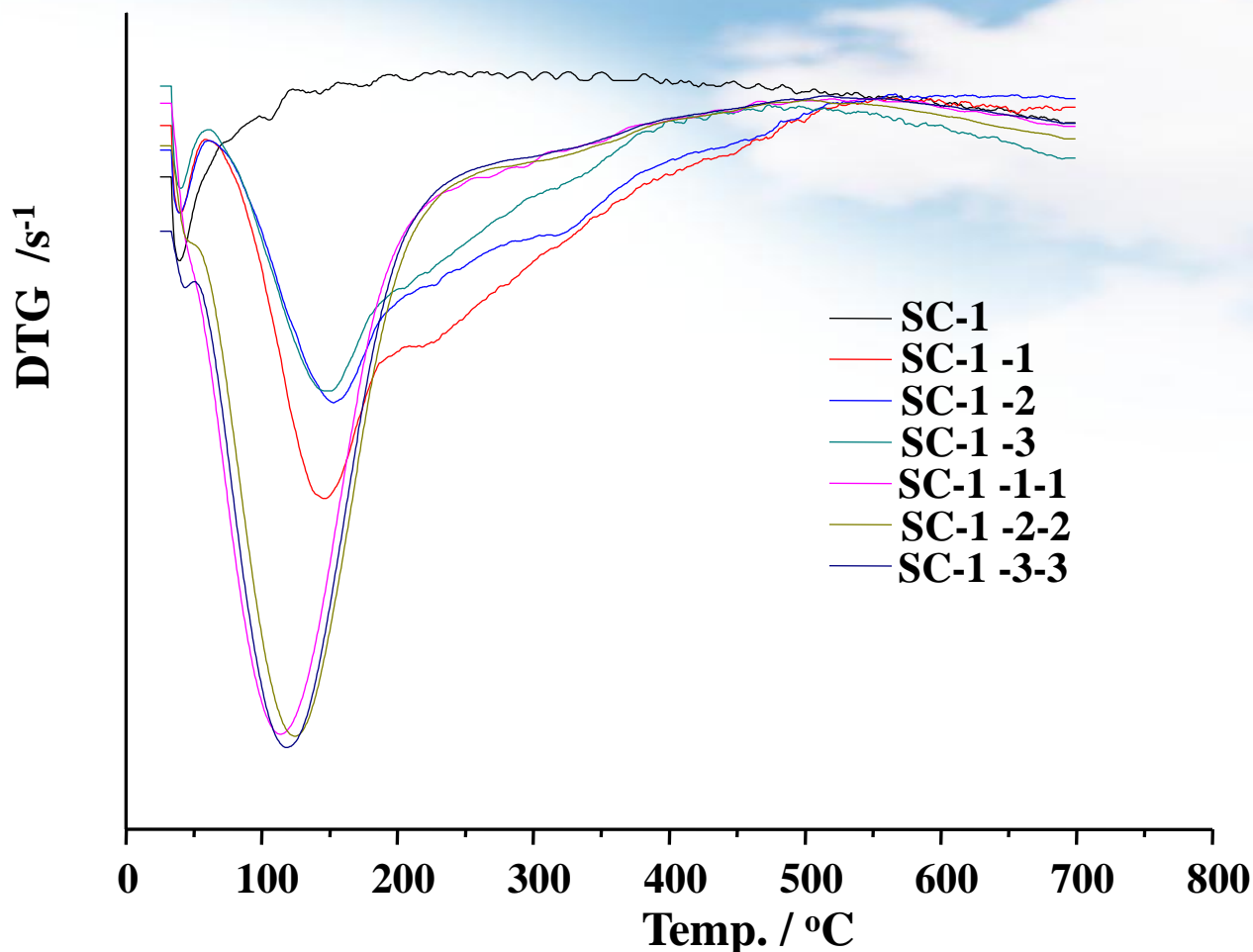
# TG curves



**TG curves of sample SC-1 series**

The porous carbon with different position in the filter showed different weight loss, indicating different absorb amount of the tar and the toxicants.

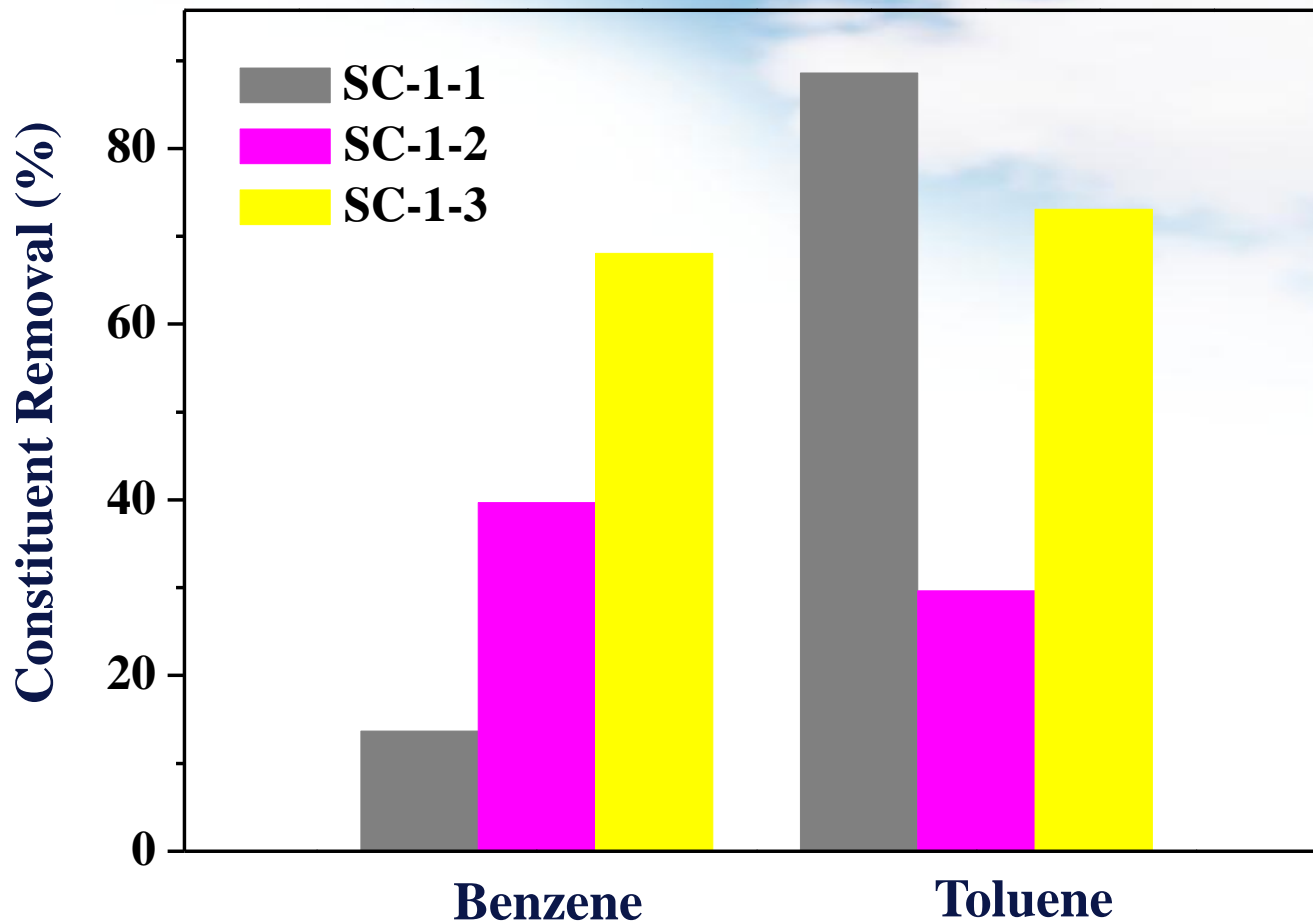
# DTG curves



**DTG curves of sample SC-1**

**A very fast mass loss was observed at ca. 110 °C. That meant most of the adsorbed compounds volatilize from carbon SC-1.**

# Removal efficiency



**Constituent removal of benzene and toluene using porous carbon SC-1**

**Aromatic compounds benzene and toluene were selected as probe molecules. Protected carbon showed better adsorption performance.**

# Conclusions

- **The deposition of tar on the surface of coconut carbon was detrimental to the adsorption of selected toxicants in the smoke vapor phase**
- **Comprehensive analysis showed that when using porous carbon as the adsorbent of cigarette filter, appropriately protecting carbon from tar deposition could achieve the best results**

# Acknowledgement



## CORESTA for Study Grant

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**THE END!**

**Thank you for your attention!**





# CORESTA CONGRESS EDINBURGH 2010

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Promoting the Scientific Basis for  
Tobacco Product Regulation

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